Phil Horwitz: A Career in Separation Science

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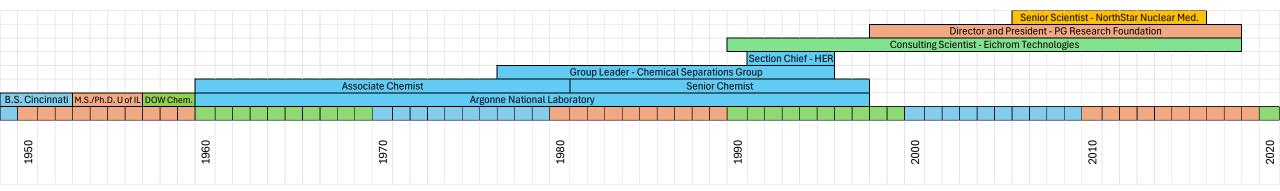
68th RRMC

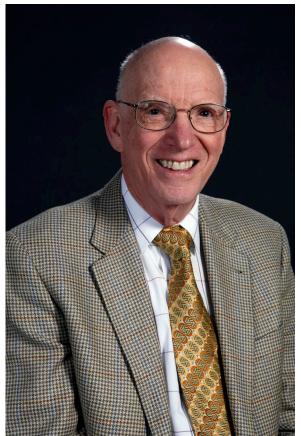
November 4, 2025

Idaho Falls, ID









Born: June 3, 1930

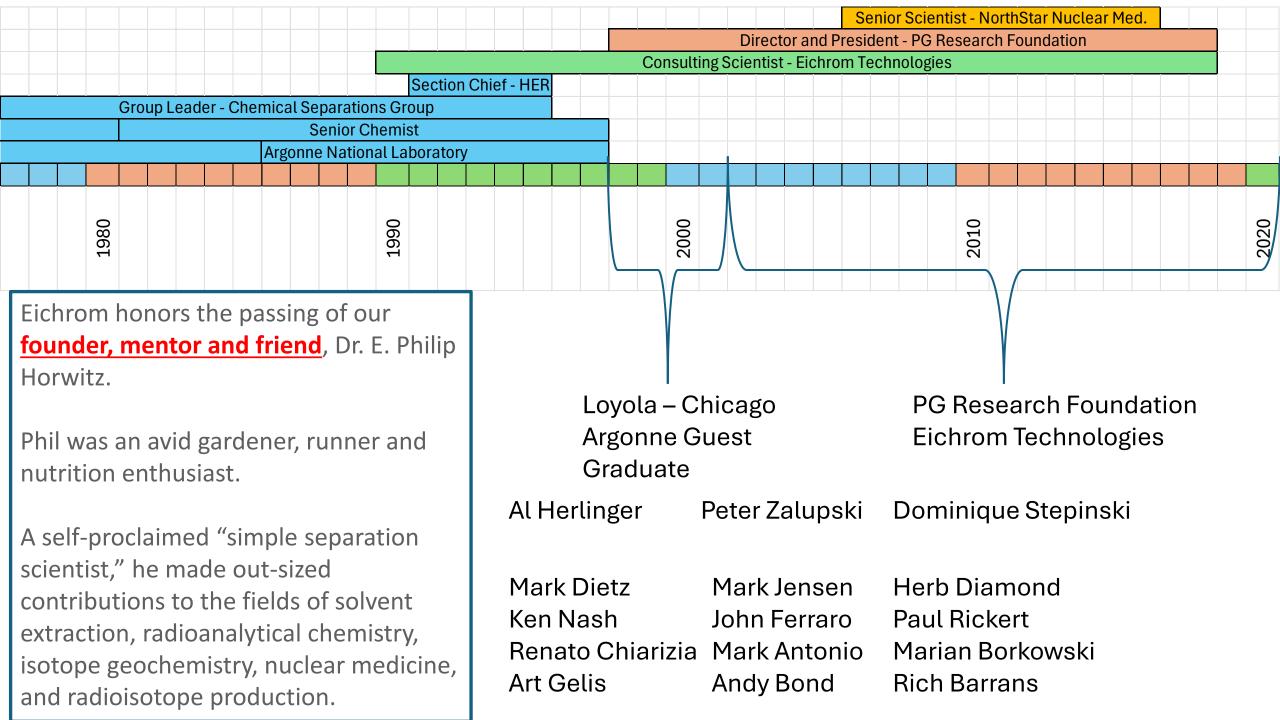
Died: December 28, 2024

Reviews of Phil Horwitz Research:

D.R. McAlister, H. Kurosaki, S. Happel, E.P. Horwitz, "Development of EXC Resins for Metal Ion Separations," Bunseki, 2, 44-50, (2020).

(Contact speaker for English language version.)

E.R. Bertelsen, J.A. Jackson, J.C. Schafer, "A Survey of Extraction Chromatographic f-Element Separations Developed by E. P. Horwitz," Solv. Extr. Ion Exch., 38(3), 251-289, (2020).



PGRF Lab



Phil's CV

45 patents, >180 peer reviewed journal articles, hundreds of conference presentations

TRUEX Process (CMPO) - Minor Actinides from PUREX Raffinate)

SREX Process (DtBCH18C6) – Sr from acidic spent nuclear fuel waste

ABEC Resin (PEG-2000) – Tc from caustic spent nuclear fuel waste and Tc-99m from low specific activity Mo-99

4 R&D 100 awards (TRUEX, EXC Resins, Diphonix, ARSIIe)

Glenn T. Seaborg Actinide Separations Award – 1992

Becquerel Medal (RSC Radiochemistry Group) – 2004

Founders Award (Radiobioassay and Radiochemical Measurements Conference) – 2007

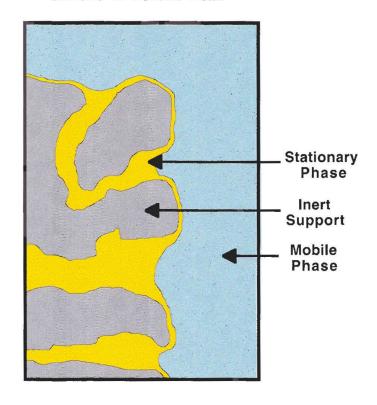
Glenn T. Seaborg Award for Nuclear Chemistry (ACS) - 2016

2016 ACS Symposium (Seaborg Award)



Depiction of Extraction Chromatography (EXC)

Surface of Porous Bead



Relationship Between SX and EXC

$$\mathbf{k}' = \mathbf{D} \bullet \frac{\mathbf{v}_{\mathbf{S}}}{\mathbf{v}_{\mathbf{m}}}$$

k' = retention volume (FCV to peak maximum)

D = distribution ratio

 v_s = volume of stationary phase

 v_m = volume of mobile phase

Lipophilic (Hydrophobic) Extractants Physically Sorbed in the pores of an inert support.

Combines selectivity of solvent extraction with the ease of operation of chromatography

Not covalently bound

Limited to aqueous mobile phases

Phil's Mantras

- 1) Chance favors the prepared mind
 - Better to be lucky than good
 - Borrowed from Louis Pasteur
 - Develop and maintain a toolbox of skills/methods
- 2) Always extractant the minor component (unless you can't, then use solvent extraction).
- 3) What goes in must come out (the solvent extraction graveyard is full of compounds with very high D, k', Dw, Dv.....)
- 4) Always plot elution curves on a semi-log scale

Phil's Mantras (cont'd)

Work safely

- No evaporations
- Smaller columns/less waste
- Faster chemistry without compromising quality

Learn Together (Collaborate)

- National labs
- Universities
- Industry

Train and learn from the next generation of scientists



Phil's Lunchtime Discussions

Any topic, often not politically correct

"I don't know" is not an acceptable answer

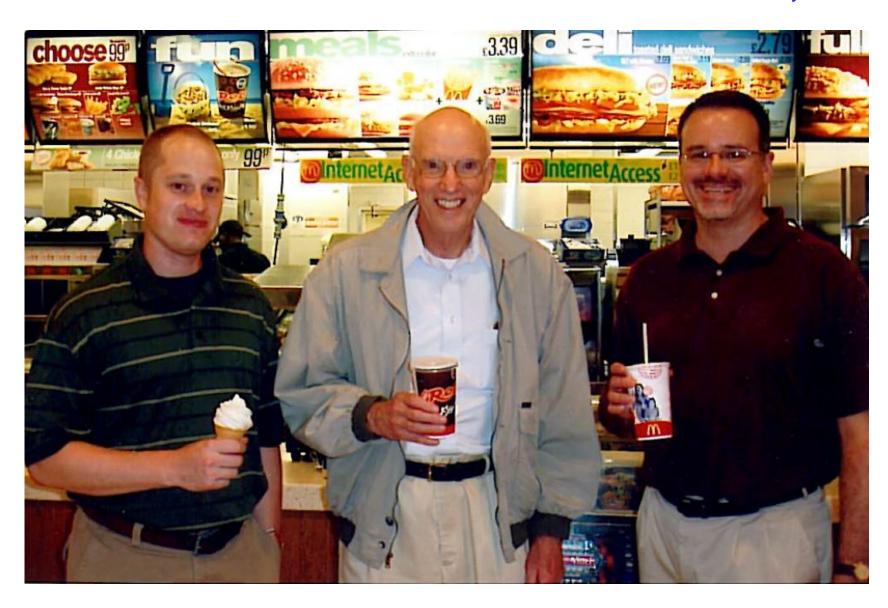
If you sit at Phil's table (or any adjacent table touching Phil's table), you must participate

Phil's lunch did not look delicious

- -proprietary nona-veggie mix
- -low sodium very veggie juice
- -juice plus vitamins

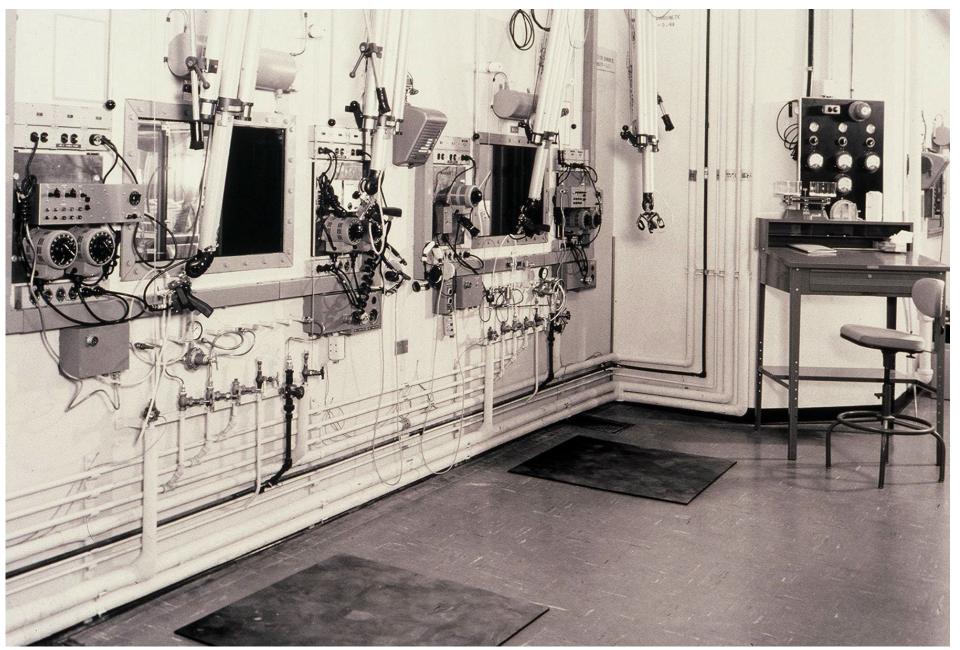


Phil didn't like to travel, but...



Dan, Phil, and Mike (Oxford McDonalds, 2006)

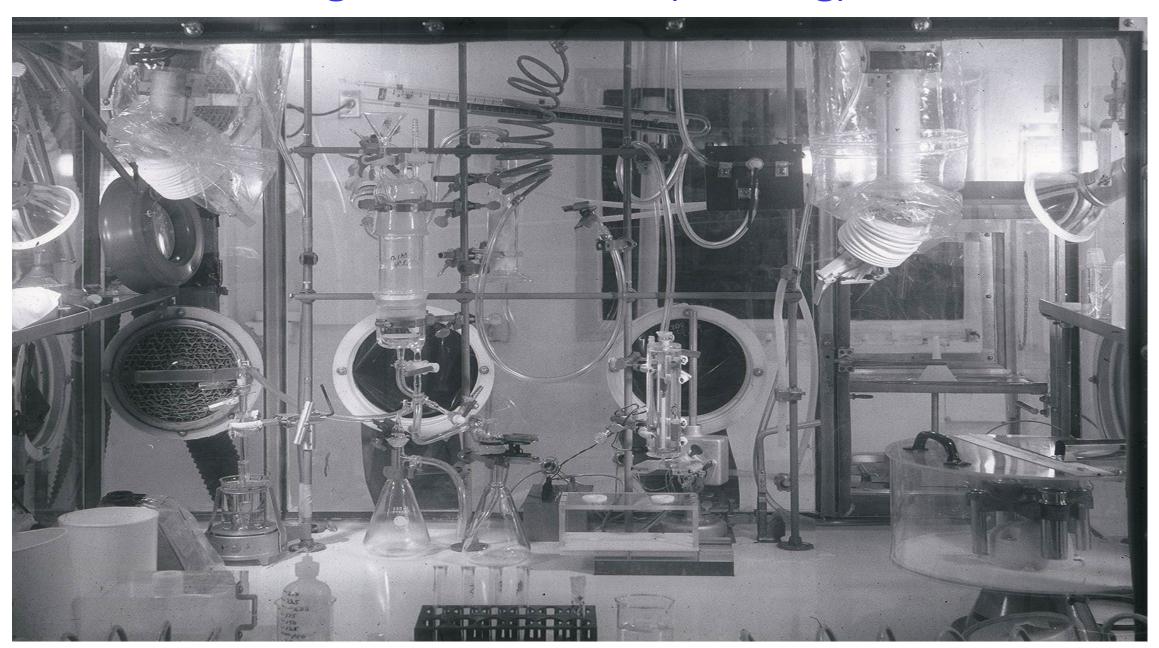
Argonne Hot Cells (M-Wing)



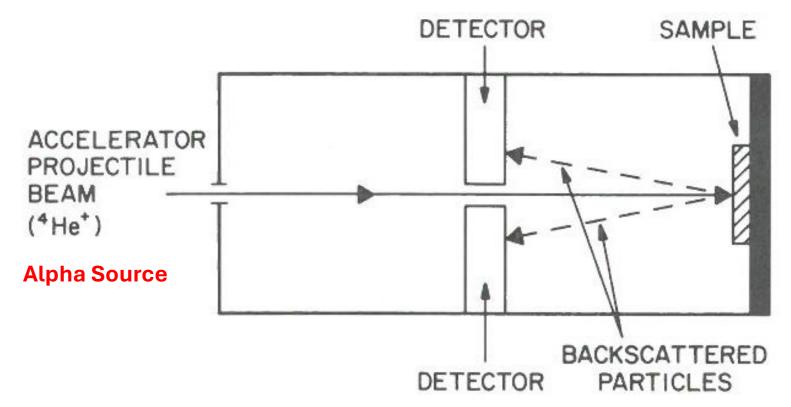
Argonne Hot Cells (M-Wing)



Argonne Hot Cells (M-Wing)



²⁴²Cm alpha sources for Lunar Exploration



Rutherford Alpha Backscatter

Energy of scattered particle = element

Intensity of each peak = relative amount

²⁴²Cm alpha sources for Lunar Exploration

Alpha Scattering Experiment of the NASA Surveyor Lunar Program

Required: point source of α particles of high intensity and quality

Criteria for selection of alpha-active isotope

- Half-life < 1 year to have weightless source
- Half-life > 2 months to minimize decay before measurements were completed
- Alpha particles have to be monoenergetic and > 5.8 MeV
- Isotope has to be available in curie quantities free of isotopic interferences
- Chemical properties has to allow separation from <u>all</u> impurities

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Preparation of 242Cm (t_{1/2} = 162.5 \text{ days})
Neutron capture of 241Am
150-250-mg samples <sup>241</sup>Am encapsulated
in high-purity aluminum
    241Am (n,\gamma) 242Am \rightarrow 242Cm + \beta 95 96
                   16% ∫ E.C.
                      242Pu
                        94
 Dates of 242Cm Production Runs
    November 7-15, 1963
    November 11-30, 1964
    September 8-27, 1966
    November 13-27, 1967
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E.P. Horwitz, C.A.A Bloomquist, H.W. Harvey, D. Cohen, L.J. Basile, "Purification of 10 Curies of 242Cm," Argonne National Laboratory Report ANL-6998, February (1965).

Dissolved in nitric acid

Evaporated

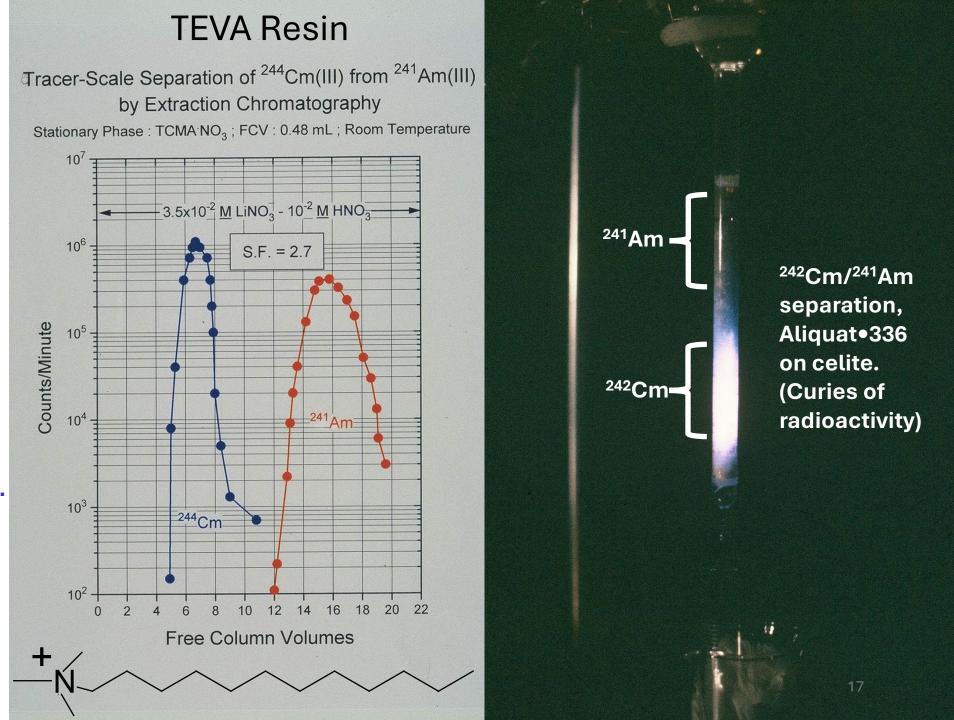
Dissolved in LiNO₃ for Am/Cm extraction by liquid anion exchanger (remove Al-cladding)

Am/Cm recovered in 3M HCl.

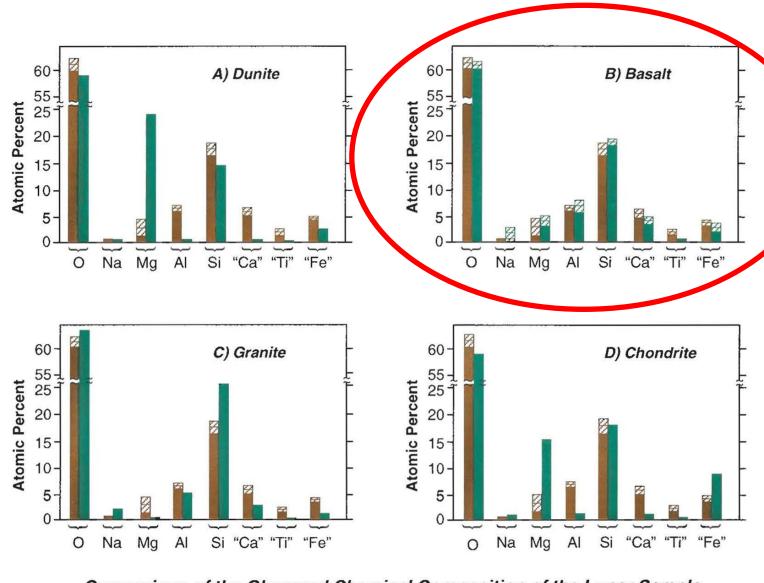
Am(V) precipitated as KAmO₂CO₃.
Cm(III) remains soluble.
(Main Am removal step).

Final Cm purification by EXC with liquid anion exchanger. (TEVA).

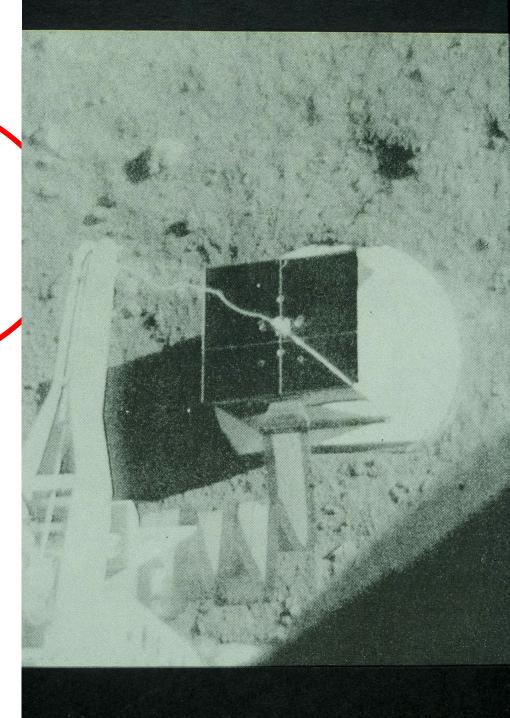
Removes final traces of Am-241.



Comparison of Lunar Surface to Known Minerals

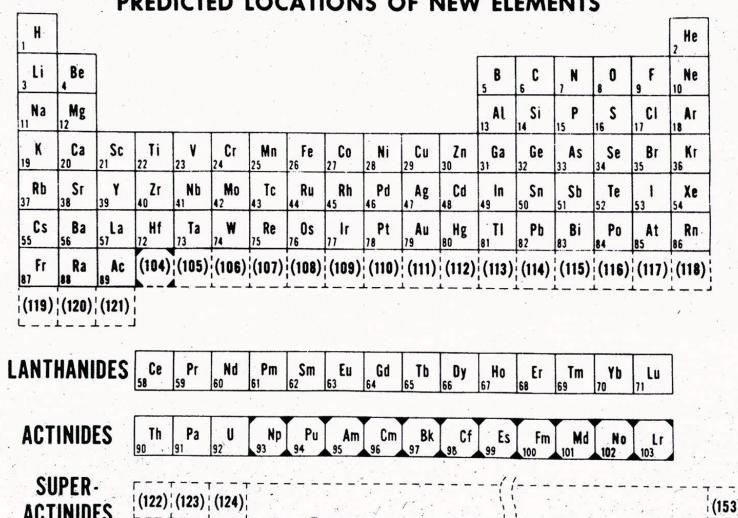


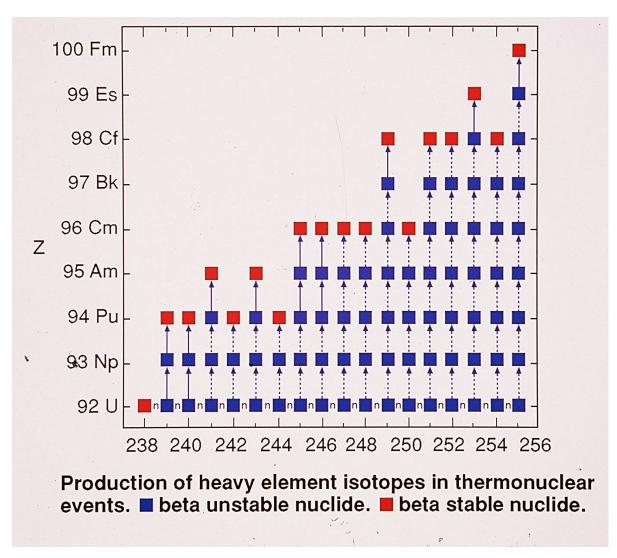
Comparison of the Observed Chemical Composition of the Lunar Sample (Brown Bars) with the Average Composition of Selected Materials (Green Bars)

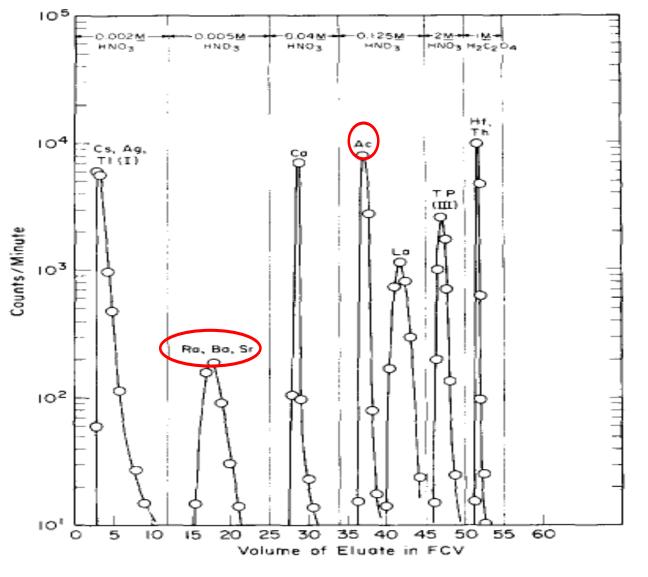


Search for new elements (1969-1971)

CONVENTIONAL FORM OF PERIODIC TABLE SHOWING PREDICTED LOCATIONS OF NEW ELEMENTS



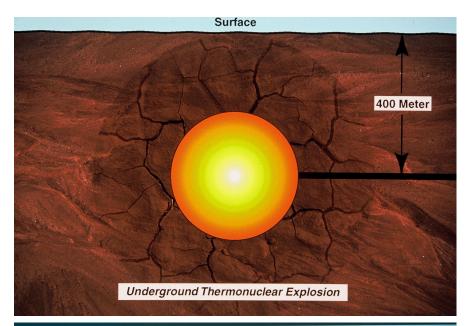


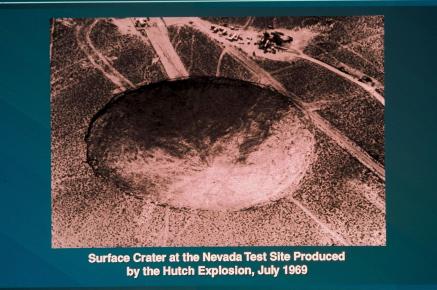


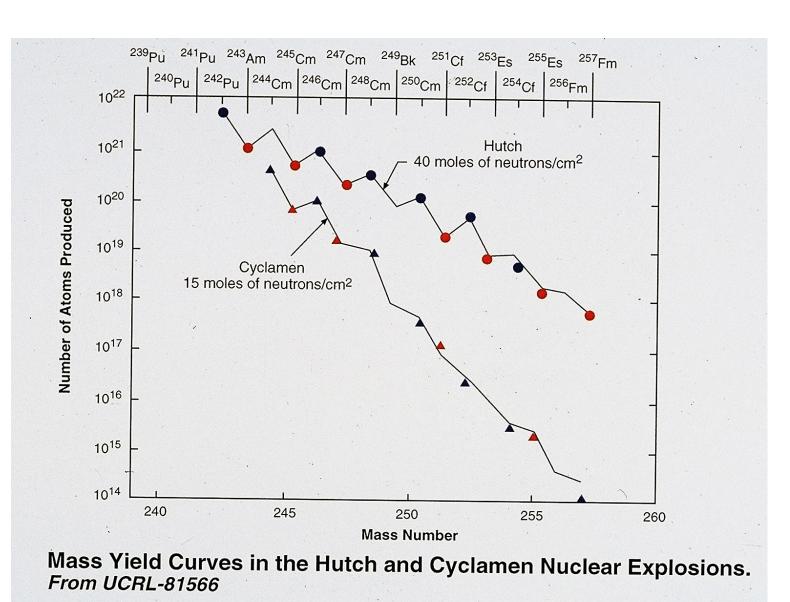
O-P-OH O

LN Resin

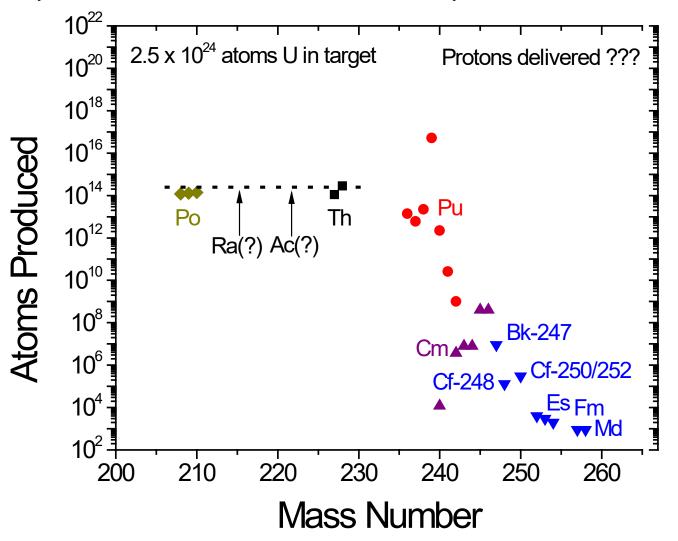
Fig. 9. Sequential separation of group C elements using 8.82 w/o HDEHP on Celite (35 μ). Column bed size $0.062 \text{ cm}^2 \times 5 \text{ cm}$; 50°C ; $v = \sim 5 \text{ cm/min}$; FCV = 0.22 ml.







Spallation Yield for Uranium Beamstop with 12 GeV Protons

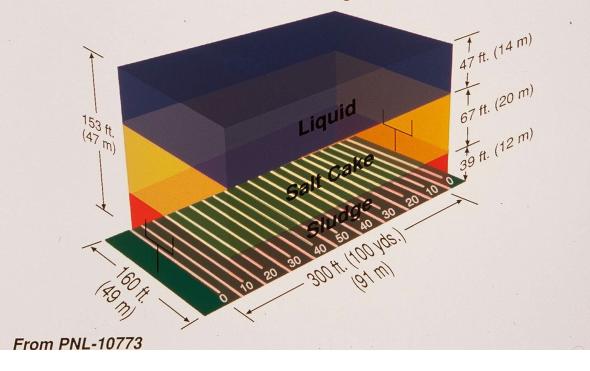


J.P. Unik, E.P. Horwtiz, K.L. Wolf, I. Ahmad, S. Fried, D. Cohen, P.R. Fields, C.A.A. Bloomquist, D.J. Henderson, "Production of Actinides and the Search for Super-Heavy Elements Using Secondary Reactions Induced by GeV Protons," *Nuclear Physics*, A191, 233-244 (1972).

Waste tank cleanup



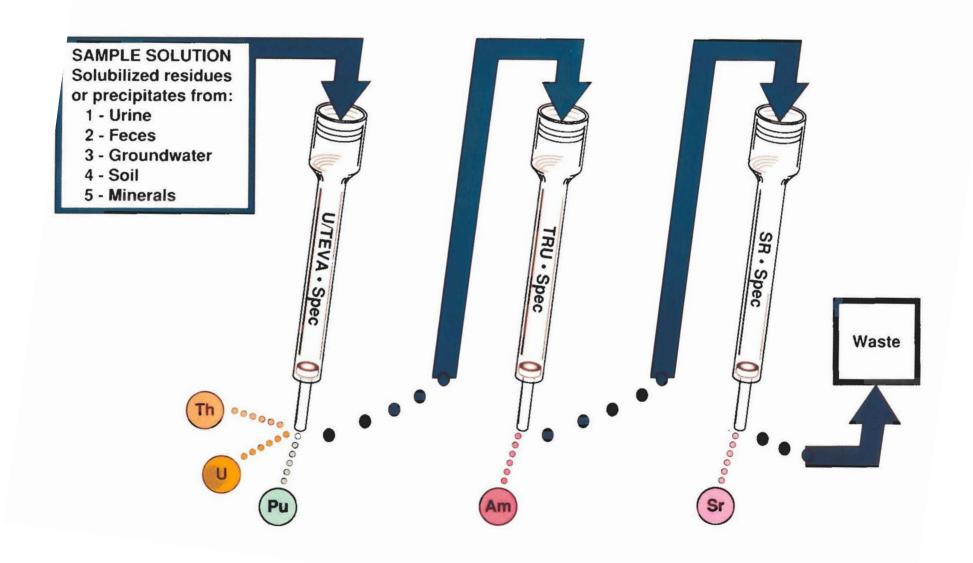
The 55 million gallons (208 x 10³ cubic meters) of radioactive waste in Hanford's underground storage tanks would fill a football field to a height of about 150 feet.



Waste tank cleanup

Process	PUREX	TRUEX	SrEx
Target	U/Pu	Am/Cm/minor actinides	Sr(Pb)
Extractant	O C ₅ H ₁₁ O C ₅ H ₁₁ O C ₅ H ₁₁ TBP (DA[AP])	CMPO/TBP	DtBCH18C6
Resin	UTEVA	TRU(RE)	Sr Resin (Pb Resin)

Multi-Radionuclide Separation using Tandem Spec Column Arrangement



Phil's Legacy (Eichrom Current R&D Team)



Daniel McAlister, Ph.D. - Group Leader
23 years at Eichrom/PGRF
Loyola University Chicago
Guest Graduate Argonne National Laboratory
Separation Science, Radiation Protection, Analytical Chemistry

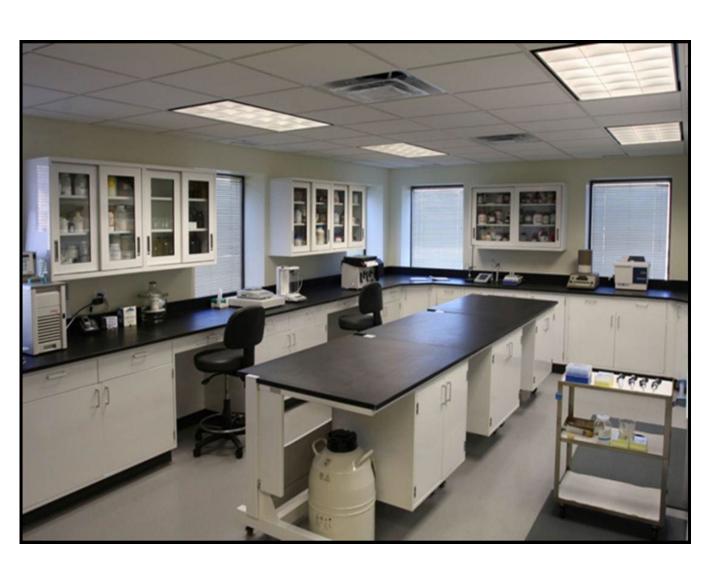


Eddie Rush, B.S. - Scientist 17 years at Eichrom University of Illinois at Chicago Organic Synthesis, Radiochemistry



Madeleine Eddy, Ph.D. - Scientist 3 years at Eichrom Colorado School of Mines Radiochemistry, Solvent Extraction, Actinide-Lanthanide Separations

Phil's Legacy (Eichrom Resources)



Radioactive Materials Laboratory

Broad Scope License, any radionuclide of elements 1(H) – 103(Lr), no Rf, Db, Sg... ⊗

Permitted mCi amounts of Ac-225, Ac-227, Pb-203, Th-228, 1Ci F-18.

Alpha spectrometry, HPGe gamma, LSC, Na(TI)I gamma, MP-AES, ICP-MS

Cyclotron

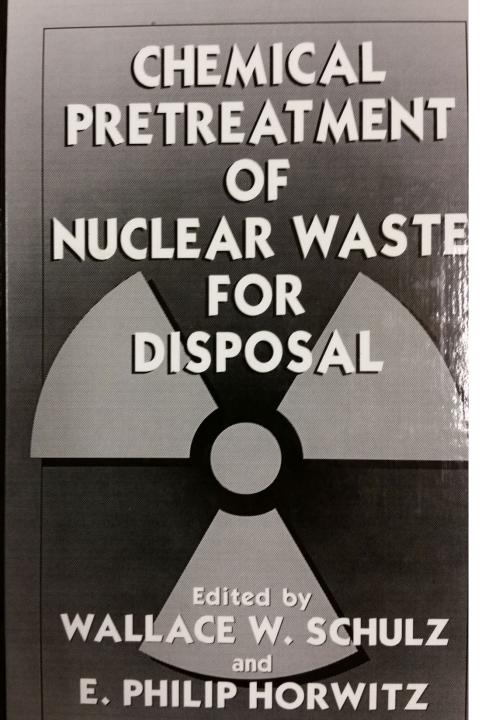
Nuclear Reactor

Phil's Legacy (Eichrom Library)



Best of good fortune to a fine Chemist and persons

Phil Horwitz



ADVENTURES IN RADIOISOTOPE RESEARCH

The Collected Papers of GEORGE HEVESY in Two Volumes

VOLUME Two

PERGAMON PRESS OXFORD · NEW YORK · LONDON · PARIS 1962 Men 3 Sauborg Apr. 1962

Slevn 3. Seaborg

https://en.wikipedia.org/wiki/Glenn_T._Seaborg



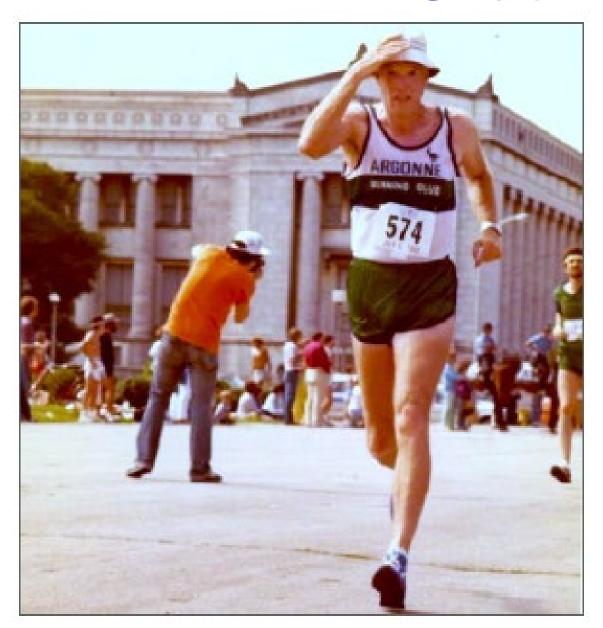
Phil was amazing both professionally and personally - had an opportunity to travel with him and found he always took the stairs when possible, grew his own vegetables and would often pass on the high end restaurants for simpler meals. He once talked me into McDonalds as we were on the way to meet a larger group for dinner, that meal was quite an honor.



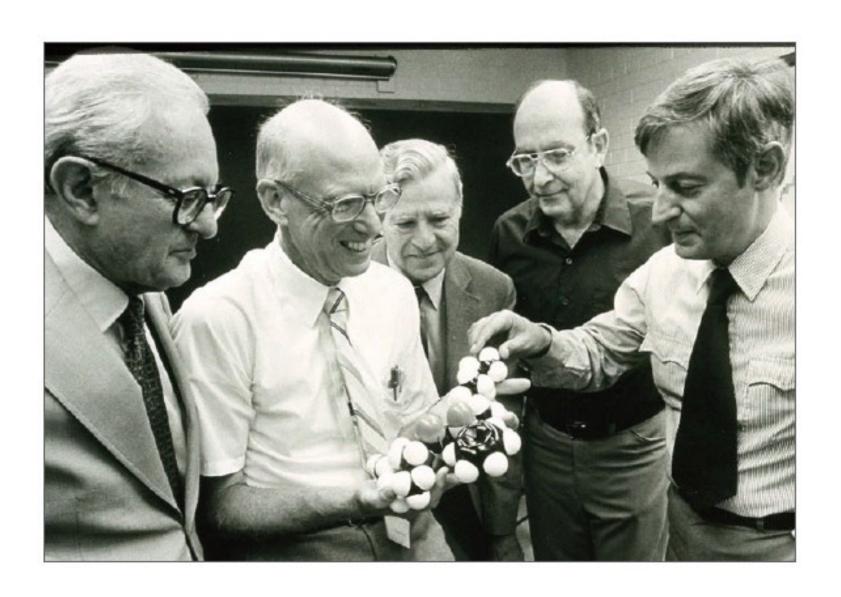
Though I never had the privilege of knowing Dr. Horwitz personally, his work has profoundly influenced my journey in the field of radiochemical separations. I began my studies by delving into his groundbreaking papers on extraction chromatographic separations, which laid the foundation for so many of us in the field. His legacy will continue to inspire generations of researchers.



Phil was inspiration to us, and he was a mentor to me. His scientific contributions are well known and as a person, he demonstrated the healthy lifestyle that was ahead of its time; as a runner in any weather and diet that was packed full of raw vegetables. 'Always firing on all cylinders' I am so fortunate to have known you. ILLINI!



Phil was a great scientist, mentor to many, a terrific human being, and an avid college football fan. He always gave me a hard time about FSU being so highly ranked in the polls as he didn't think the ACC was much of a conference compared to the Big-10 (Big-18 now?). After this year, I am starting to agree with him...



Phil understood the importance of solvent extraction, often with a smile referring to it as the "premier technique." He followed up his conviction by co-founding the journal Solvent Extraction and Ion Exchange in 1983, serving as Editor in Chief for the next two decades.



I doubt that the world will ever understand or even know of the beneficial changes introduced by Phil to the field of analytical chemistry. Those of us who benefited from those enhancements will be forever grateful.



If it weren't for Phil, radiochemists couldn't do much of what we do the way we do it. It was ~33 years ago when I first used SrSpec columns that allowed me to do Sr seperations in the back of a semi trailer out on a field site.



It's hard to imagine my career without the time at Eichrom. It was there that I really started to become an independent scientist. Phil's creativity and simplicity in presenting ideas left a mark that has followed throughout my career.

Mike Fern 35 years at Eichrom

