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Rapid Determination of Pu in Steel Samples

Figure 1. Sample Preparation

Add 1-2 g steel sample to Teflon beaker*.

AN-1802-10

Summary of Method Plutonium is separated and measured from 1-2 gram steel samples. Samples are digested with concentrated nitric, hydrochloric, and hydrofluoric acids. The digestate is evaporated to dryness, the residue dissolved in HNO₃/H₃BO₃, and a CaF₂/LaF₃ precipitate is used to concentrate the Pu and remove matrix. An optional NaOH fusion may also be performed, post sample digestion, to dissolve concrete or stone included in the sample and deal more rigorously with refractory Pu. Plutonium is separated from matrix impurities and potentially interfering radionuclides in the sample using 2 mL cartridges of Eichrom TEVA Resin. Plutonium is measured by alpha spectrometry following rare earth fluoride microprecipitation onto Eichrom Resolve filters. The chemical recovery of Pu, determined by ²⁴²Pu tracer, was 90–99%. Measured values of Pu typically agreed to within 7-8% of reference values for 16 hour count times. The minimum detectable activity for Pu in 2 g samples with 16 hour count times was 0.25 mBq/g.

A single operator can prepare batches of 12 samples for the measurement of Pu in less than 8 hours.

Reagents

Muffle Furnace*

Analytical Balance

Teflon Beakers (Zr Crucibles*)

Alpha Spectrometry System

50 mL and 250 mL Centrifuge Tubes

Yellow Outer Tips (Eichrom AR-1000-OT)

Vacuum Box (Eichrom AR-24-BOX or AR-12-BOX)

Cartridge Reservoir, 20 mL (Eichrom AR-200-RV20)

Inner Support Tubes-PE (Eichrom AR-1000-TUBE-PE)

Resolve Filters in Funnel (Eichrom RF-DF25-25PP01)

Hot Plate

If using optional fusion, omit HF/H₃BO₃ and use Zr crucible. TEVA Resin, 2 mL Cartridges (Eichrom TE-R50-S) Nitric Acid (70%) Add ²⁴²Pu tracer, 5 mL 70% HNO₃, Hydrochloric Acid (37%) Hydrofluoric Acid (49%) or Ammonium Bifluoride 20 mL 37% HCl, and 5 mL 49% HF. Lanthanum Carrier (10 mg/mL) Digest on Hotplate to dryness. Cerium Carrier (10 mg/mL) Add 1 mL 70% HNO3. Deionized Water 1.25M Ca(NO₃)₂ Optional NaOH Fusion. ²⁴²Pu Tracer 10 mL 37% HCl, and 1 mL 49% HF. 2M AI(NO₃)₃ Add 15-20 g NaOH. Boric acid Digest on Hotplate to dryness. NaNO₂ Fuse at 600C for 20 min. Ascorbic Acid 30% H₂O₂ 10-20% (w:w) TiCl₃ in HCl Add 5 mL 3M HNO₃-0.25M H₃BO₃ Dissolve in DI water. 3.2M (NH₄)₂HPO₄ and 5 mL 37% HCl. Transfer to 250 mL c-tube. Sodium Hydroxide* Digest on Hotplate to dryness. Dilute to 100 mL. Add 2 mL 1.25M Ca(NO₃)₂, 5 mg La, and Add 25 mL 1M HCl. Warm to dissolve. Equipment 8.5 mL 3.2 M (NH₄)₂HPO₄. Transfer to 250 mL centrifuge tube. Vacuum Pump Repeat 2 additional times. Centrifuge

Centrifuge. Decant _____ supernate. Dissolve ppt. in 80 mL 1.5M HCI.

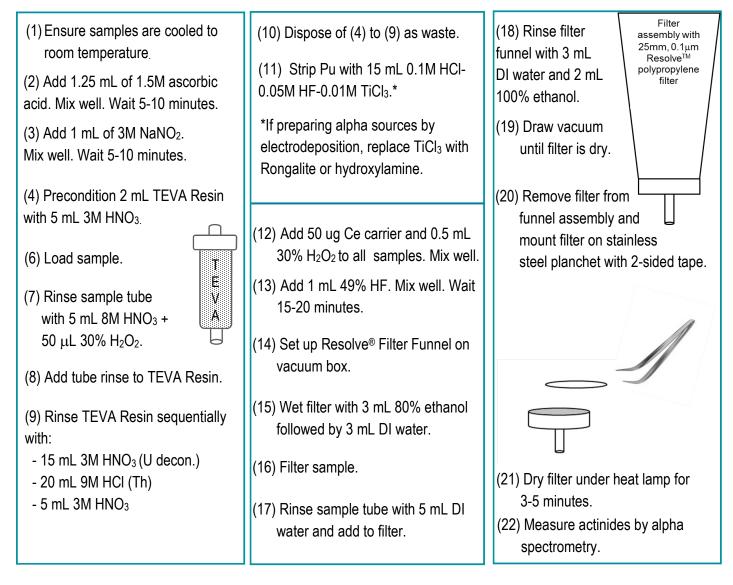
Dilute to 170 mL with 0.01M HCl. Add 2 mL 1.25M Ca(NO3)₂,5 mg La, and 3 mL 20% TiCl₃.

Add 25 mL 49% HF. Mix Well. Allow to sit 20 min. Centrifuge. Discard supernate.

Dissolve precipitate in 7 mL 3M HNO₃-0.25M H₃BO₃, 6 mL 8M HNO₃, 7 mL 2M Al(NO₃)₃. Mix. Warm if necessary. Centrifuge. Check for solids.

Continue to Pu Separation.

Figure 2. Load Solution Preparation and Plutonium Separation



Method Performance for Pu in Steel Samples

Details	Sample replicates	Reference (mBq/sample)	Measured (mBq/sample)	Average % Diff.	242Pu tracer % Yield
238Pu in 2 g Steel	5	37.0	37.7 <u>+</u> 1.6	4.2	89.3 <u>+</u> 2.3
239Pu in 2 g Steel	5	24.5	24.4 <u>+</u> 1.6	6.6	96.5 <u>+</u> 3.4
239Pu (refractory) in 2 g Steel	5	24.5	23.4 <u>+</u> 0.9	3.8	98.9 <u>+</u> 6.6
239 Pu in 5 g Steel	4	37.0	38.3 <u>+</u> 1.0	2.6	92 <u>+</u> 14

References

1) Sherrod L. Maxwell, Brian K. Culligan, Jay B. Hutchison, Robin. C. Utsey, Ralf Sudowe, Daniel R. McAlister, "Rapid method to determine plutonium isotopes in steel samples," *J. Radioanal. Nucl. Chem.*, 314(2), 1103-1111 (2017).