

New Chromatographic Materials for Determinations of Actinides,  
Strontium, and Technetium in Environmental, Bioassay,  
and Nuclear Waste Samples\*

E. Philip Horwitz

Chemistry Division, Argonne National Laboratory, Argonne, IL

\*Work performed under the auspices of the Office of Basic Energy Sciences, Division of Chemical Sciences, U.S. Department of Energy, under contract number W-31-109-ENG-38.

## **Co-workers**

### **Chemical Separations Group, Chemistry Division**

**Mark Dietz**

**Renato Chiarizia (Visiting Scientist, ENEA)**

**Herbert Diamond**

**Cary Bauer (Student, Northern ILL Univ.)**

### **Internal Dosimetry Group, Environment, Safety, and Health Division**

**Donald Nelson**

**Theresa Sullivan**

### **Analytical Chemistry Group, Chemical Technology Division**

**Lesa Smith**

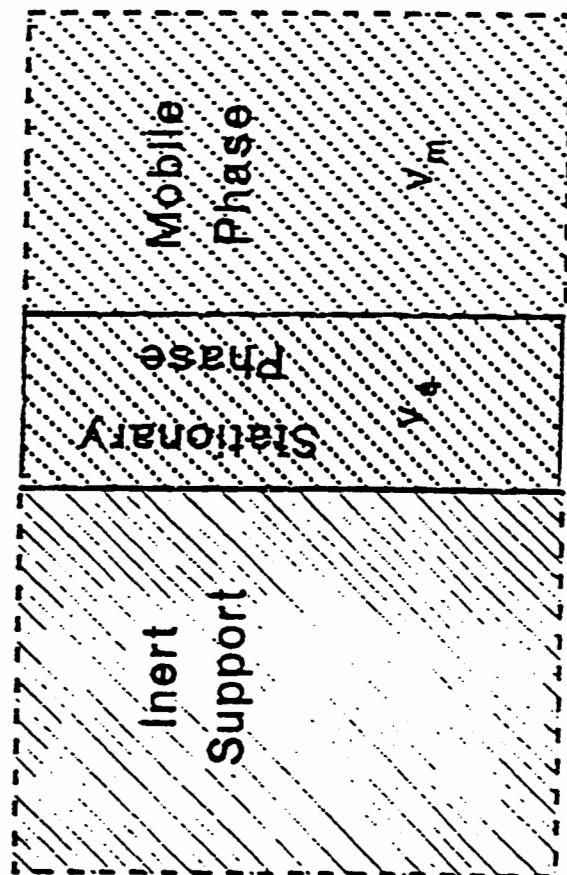
## TABLE OF CONTENTS

### FUNDAMENTALS

A.	Schematic drawing of extraction chromatographic system .....	1
B.	Features of the new chromatographic materials.....	2
C.	Currently available extraction chromatographic materials.....	3
D.	Characteristics of extraction chromatographic materials and columns.....	4
E.	Relationship between solvent extraction (SX) and liquid-liquid chromatography (LLC) .....	5
F.	<b>TEVA•Spec</b>	
1.	Basic Chemistry.....	6
2.	$k'$ of actinides III, IV, V, and VI vs. $HNO_3$ and HCl.....	7
3.	Effect of matrix constituents (anions) on Np(IV) retention .....	8
4.	Effect of matrix constituents (uranyl and thorium nitrates) on Np(IV) retention .....	9
5.	Separation of U(VI), Th(IV), and Np(IV) .....	10
6.	Separation of Np and Pu .....	11
G.	<b>U/TEVA•Spec</b>	
1.	Basic Chemistry.....	12
2.	Characteristics of extraction chromatographic material and packed columns .....	13
3.	$k'$ of actinides III, IV, and VI vs. $HNO_3$ .....	14
4.	Elution behavior of selected elements.....	15
5.	Effect of matrix constituents (anions) on U(VI) retention .....	16
6.	Effect of matrix constituents (anions) on Np(IV) retention .....	17
H.	<b>TRU•Spec</b>	
1.	Basic Chemistry.....	18
2.	Characteristics of TRU•Spec extraction chromatographic material and packed columns .....	18a
2.	$k'$ of actinides III, IV, V, and VI vs. $HNO_3$ .....	19
3.	$k'$ of selected ions vs. $HNO_3$ .....	20
4.	Effect of matrix constituents on Am(III) retention from 2 M $HNO_3$ .....	21
5.	Effect of Fe(II) and (III) on Am retention from 2 M $HNO_3$ .....	22
6.	Effect of matrix constituents on Np(IV) retention from 2 M $HNO_3$ .....	23
7.	Elution behavior of selected elements.....	24
8.	$k'$ of actinides III, IV, and VI vs. HCl.....	25
9.	Effect of oxalic acid on U(VI) and Np(IV) retention from 1 M HCl .....	26
10.	Sequential separation of actinides .....	27
I.	<b>Sr•Spec</b>	
1.	Basic Chemistry.....	28
2.	Characteristics of extraction chromatographic material and packed columns.....	29
3.	$k'$ of alkali and alkaline earths vs. $HNO_3$ .....	30
4.	$k'$ of actinides and selected ions vs. $HNO_3$ .....	31
5.	Effect of matrix constituents on Sr retention from 3 M $HNO_3$ .....	32
6.	Comparison of Sr elution curves for medium and small particle sizes.....	33
7.	Elution behavior of common elements and fission products on Sr•Spec.....	34

# **FUNDAMENTALS**

# EXTRACTION CHROMATOGRAPHY



$$k' = D v_s / v_m$$

$k'$  = free column volumes to peak maximum

## **Features of the New Chromatographic Materials**

- **High specificity**
- **Room temperature operation**
- **Gravity flow**
- **Acid - side loading from a wide range of concentrations**
- **Minimal waste generation**
- **Column compatibility for tandem arrangements**

# Extraction Chromatographic Materials

3

## Currently Available

Material	Specificity
TRU•Spec	Actinides (III, IV, V), Ln(III)
U/TEVA•Spec	U(VI), Actinides (IV)
TEVA•Spec	Actinides (IV), Tc(VII)
Sr•Spec	Sr

## Currently Under Development

Tc•Spec	Tc
Ra•Spec	Ra

**Characteristics of Extraction Chromatographic  
Materials and Columns**

**Stationary Phase**      **Neat Extractant or Extractant - Diluent**

**Support**

Ambertite XAD - 7

**Particle Sizes**

50 - 100  $\mu$ m, 100 - 125  $\mu$ m

**Extractant Loading**

40 weight percent

**Bed Density**

0.33 to 0.37 g/mL

**Fractional Volumes:**

Void	0.65
Support	0.20
Extractant	0.15

## Relationship Between SX and LLC

$$k' = D \cdot \frac{v_s}{v_m}$$

where

$k'$  = retention volume

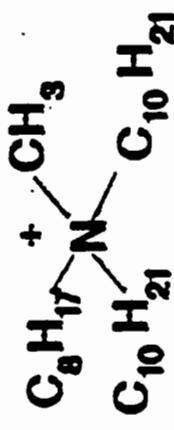
D = distribution ratio

$v_s$  = volume of stationary phase

$v_m$  = volume of mobile phase

TEVA-SPEC

## Extractant:

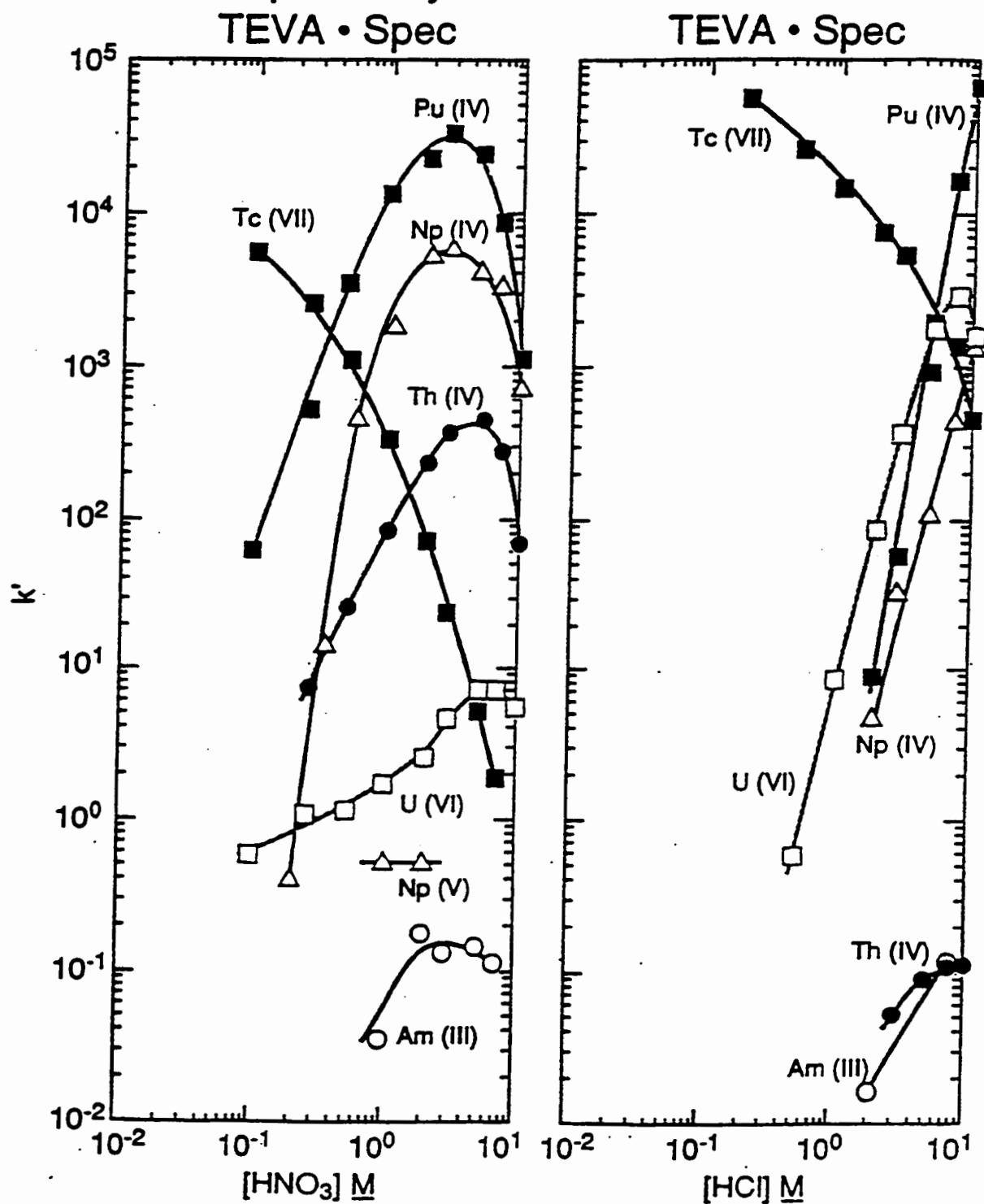


Diluent: None

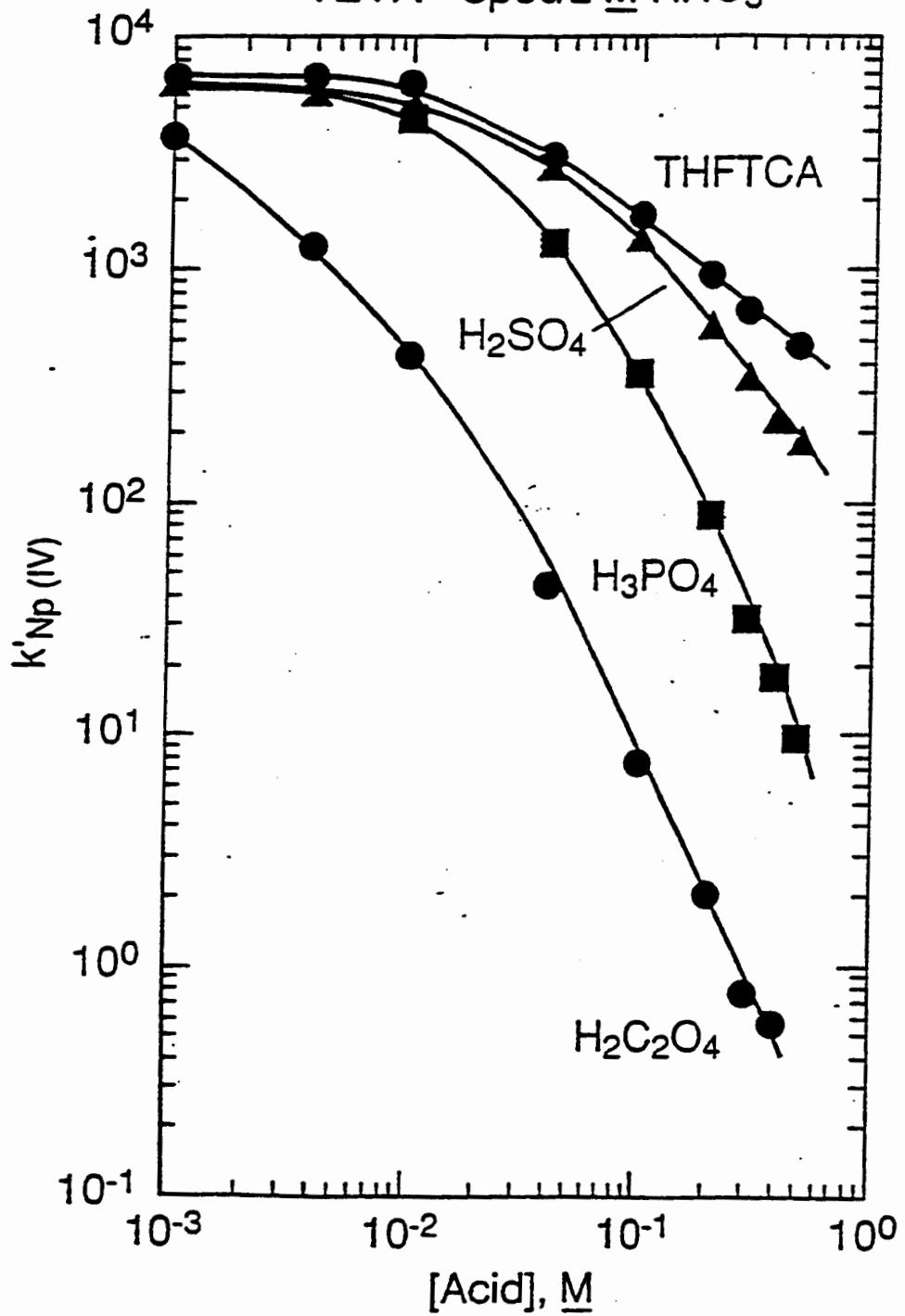
## **Extraction Equilibrium:**



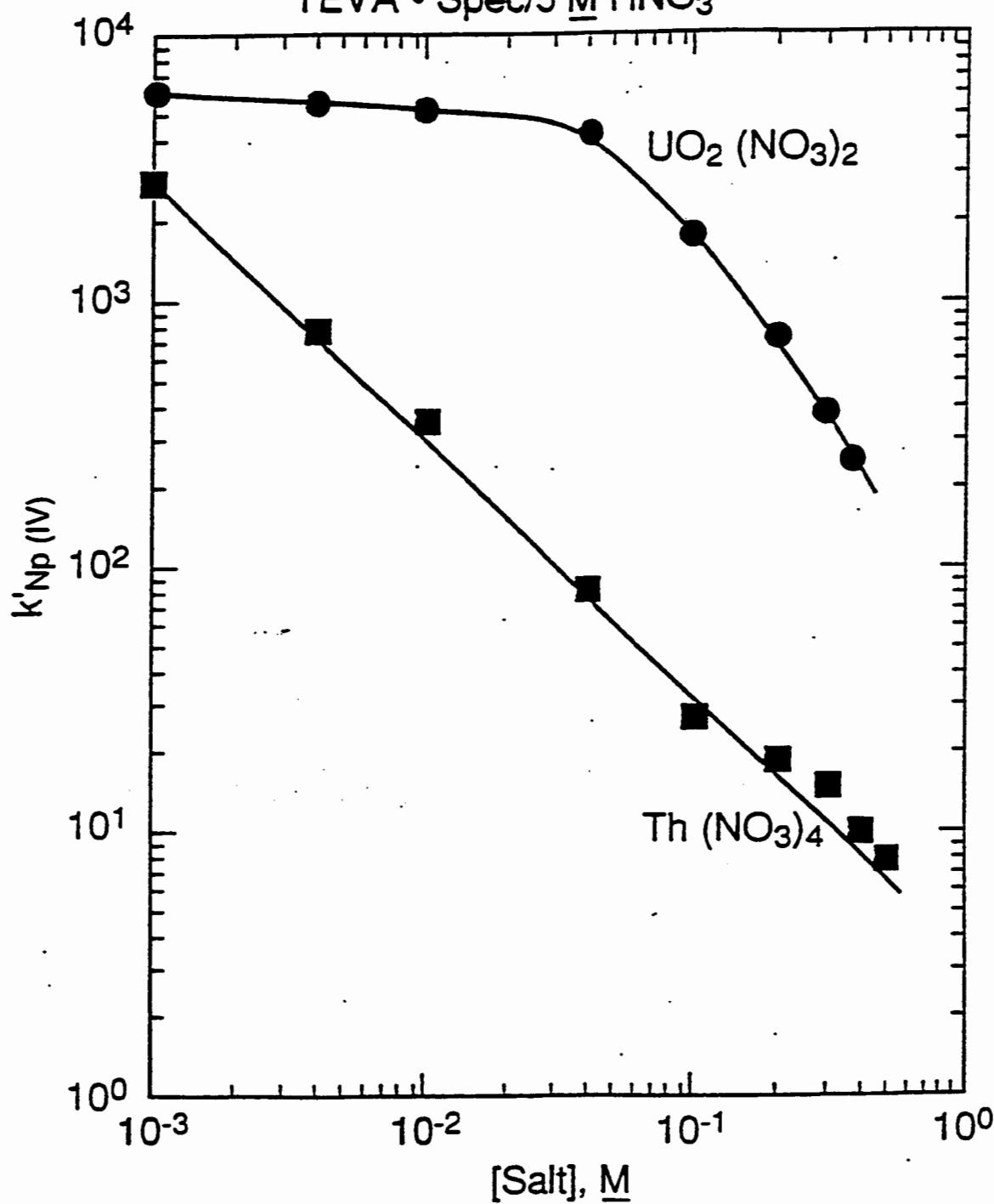
Acid dependency of  $k'$  for various ions at 23°C.

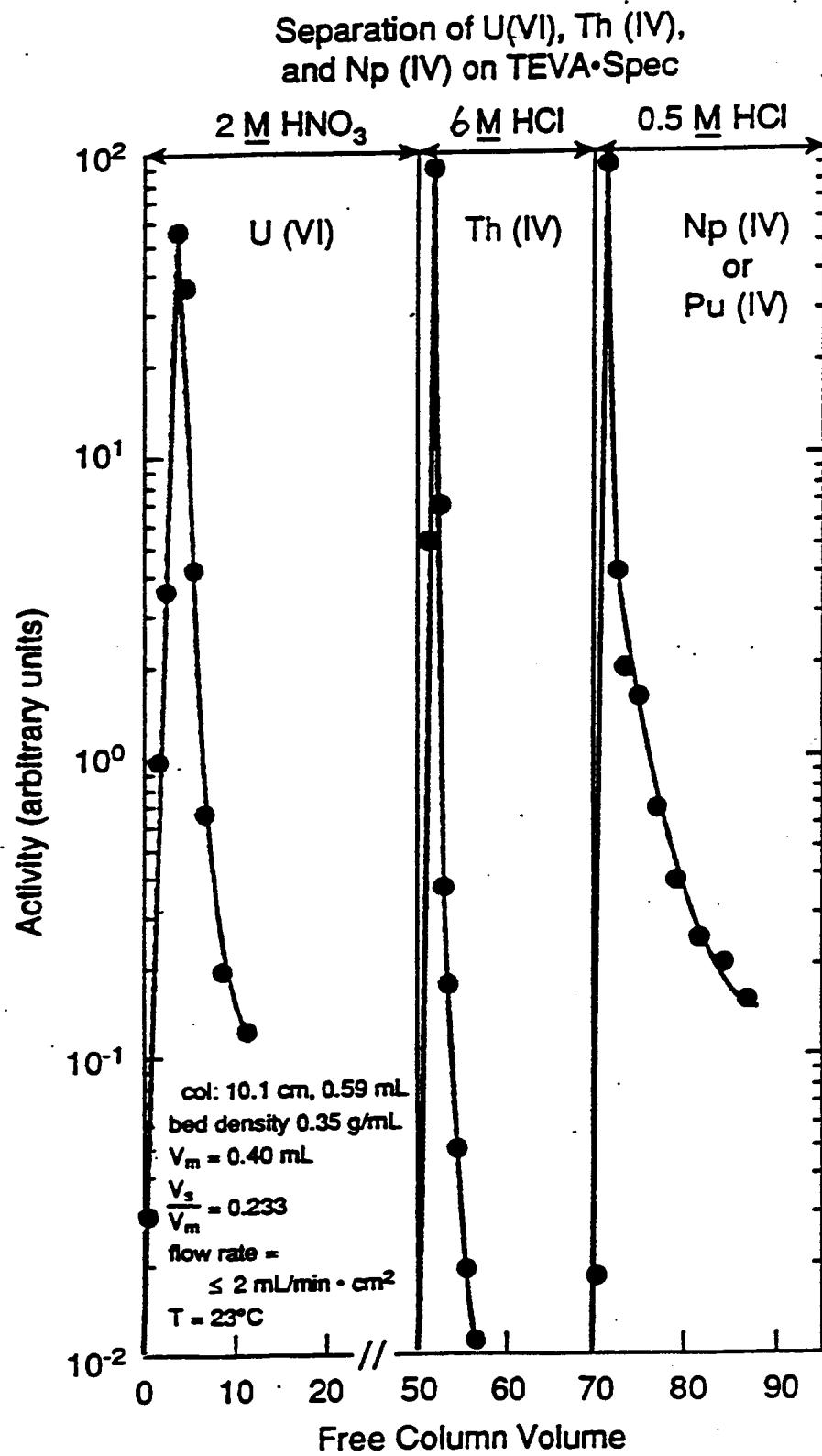


Effect of Matrix Constituents on Neptunium Retention  
TEVA • Spec/2 M HNO<sub>3</sub>



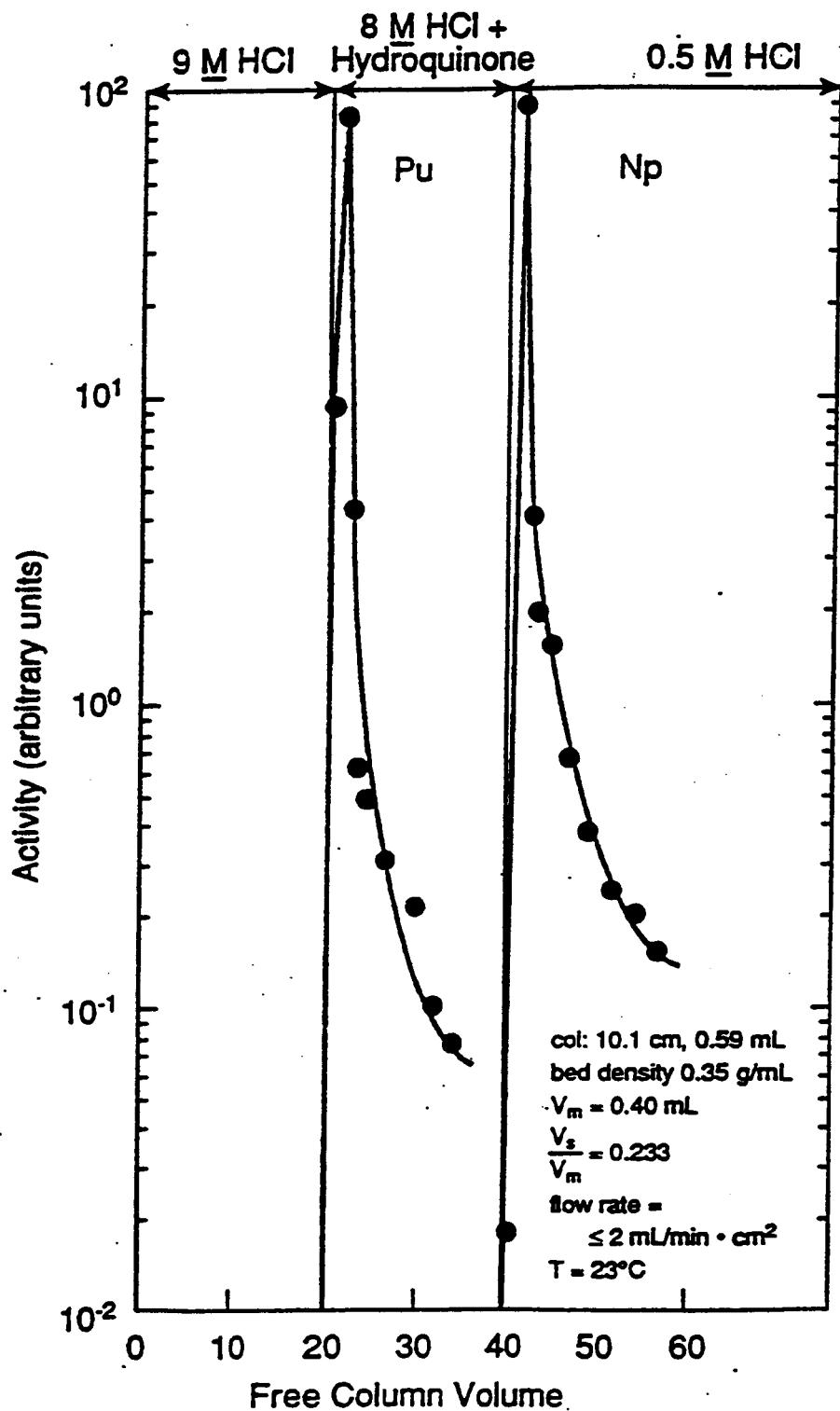
Effect of Matrix Constituents on Neptunium Retention  
TEVA • Spec/5 M HNO<sub>3</sub>



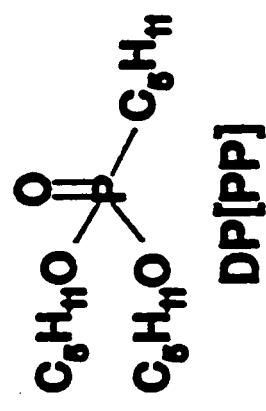
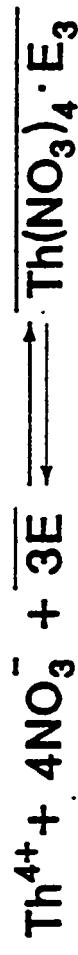


11

Separation of Np and Pu on TEVA · Spec



## U/TEVA-Spec

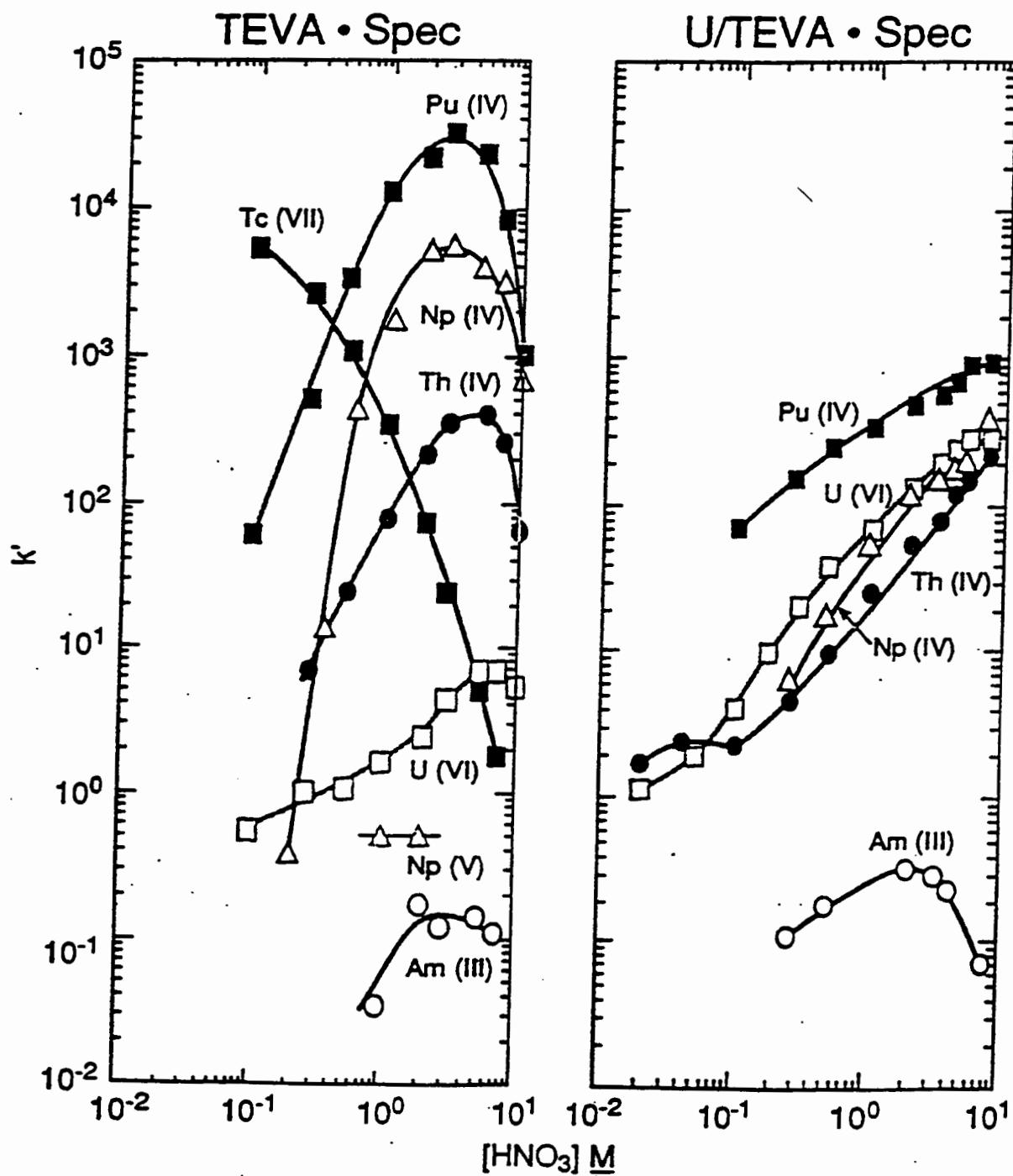
**Extractant:****Diluent:** None**Extraction Equilibria:**

**Characteristics of the U/TEVA•Spec extraction chromatographic material and packed columns.**

Bulk Material	Packed Columns
Stationary Phase	Diamyl amylphosphonate ( $\rho = 0.926 \text{ g/ml}$ )
Support	Amberchrom™ CG-71 or Amberlite™ XAD-7
Particle Diameter	50-100 $\mu\text{m}$ (Amberchrom); 80-160 $\mu\text{m}$ ; 100-125 $\mu\text{m}$
Extractant Loading	40 weight percent
Average Density of Extractant-Loaded Beads*	1.10 g/mL
<hr/>	
$v_s$	0.167 mL/mL of bed
Bed Density	0.386 g/mL
$v_m$ (also FCV)	0.65 mL/mL of bed
$v_s/v_m$	0.257
<hr/>	
Calculated Capacity	37 mg U/mL of bed
<u>Experimentally Measured Capacity</u>	37 mg U/mL of bed

\* Picnometric density and flotation density were 1.065 and 1.133, using the 50-100  $\mu\text{m}$  particle size resin. The calculated density is 1.094 g/ml assuming 100% pore filling and no swelling.

Acid dependency of  $k'$  for various ions at 23°C.



## Elution behavior of selected elements on a U/TEVA-Sotec (fine particle) column.

Element	Portion eluting (%) <sup>a</sup>						
	2 M HNO <sub>3</sub>				> 0.02 M HNO <sub>3</sub>		
Number of free column volumes <sup>b</sup>							
Element	1-5	6-10	11-15	16-20	21-25	26-30	31-40
Li	99.4	—	—	—	—	—	—
Na	95.0	<1.2	—	—	—	—	—
Mg	101	—	—	—	—	—	—
Al	92.0	<5.8	—	—	—	—	—
K	<123	—	—	—	—	—	—
Ca	92.4	<13.3	—	—	—	—	—
Cr	94.7	<3.3	—	—	—	—	—
Mn	96.5	<0.7	—	—	—	—	—
Fe	94.0	<0.3	—	—	—	—	—
Co	99.1	<7.1	—	—	—	—	—
Ni	97.8	<2.0	—	—	—	—	—
Cu	98.1	<2.6	—	—	—	—	—
Zn	96.8	<1.2	—	—	—	—	—
Rb <sup>c</sup>	102	—	—	—	—	—	—
Sr	101	—	—	—	—	—	—
Y	97.9	<2.9	—	—	—	—	—
Zr	56.4	35.7	(1.4)	—	—	—	—
Ru	59.6	<19.2	<19.2	—	—	—	—
Rh	91.7	—	—	—	—	—	—
Ag	103	—	—	—	—	—	—
Cd	96.4	<0.7	—	—	—	—	—
Cs <sup>d</sup>	100	—	—	—	—	—	—
Ba	96.6	<14.4	—	—	—	—	—
La	108	—	—	—	—	—	—
Ce	96.3	<25	—	—	—	—	—
Pr	(136)	—	—	—	—	—	—
Nd	99.0	—	—	—	—	—	—
Sm	120	—	—	—	—	—	—
Eu	100	—	—	—	—	—	—
Pb	98.4	<5.8	—	—	—	—	—
U <sup>e</sup>	—	—	—	—	—	—	>99

<sup>a</sup> Because of uncertainties inherent in the ICP-AES method used for quantitation, the fractions shown for a given element may not total 100%. Values in parentheses are subject to considerable uncertainty and are intended only as a guide. Feed solution contained ~0.02 M oxalic acid to solubilize zirconium.

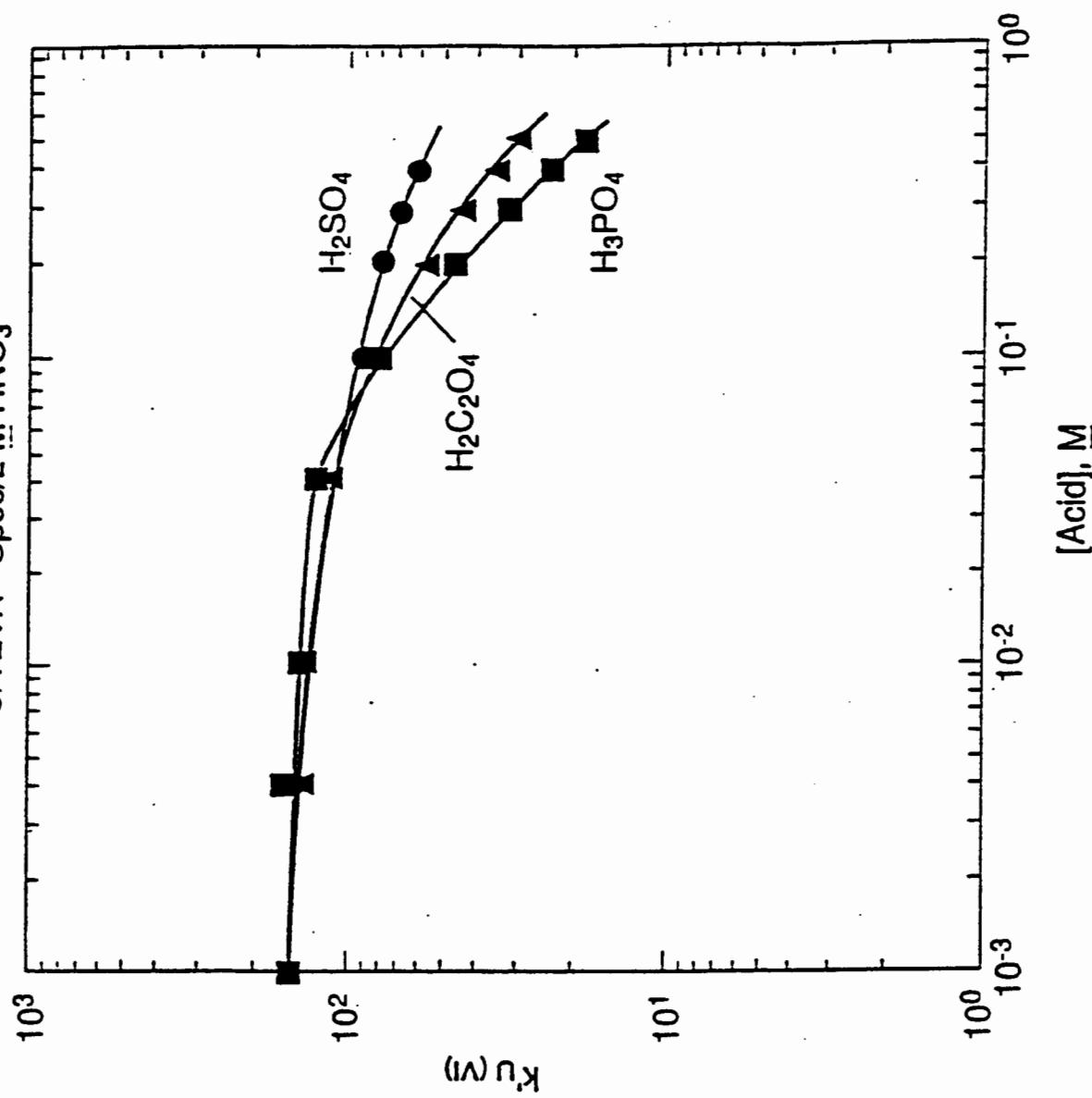
<sup>b</sup> 1 FCV = 0.60 mL here.

<sup>c</sup> By flame atomic emission.

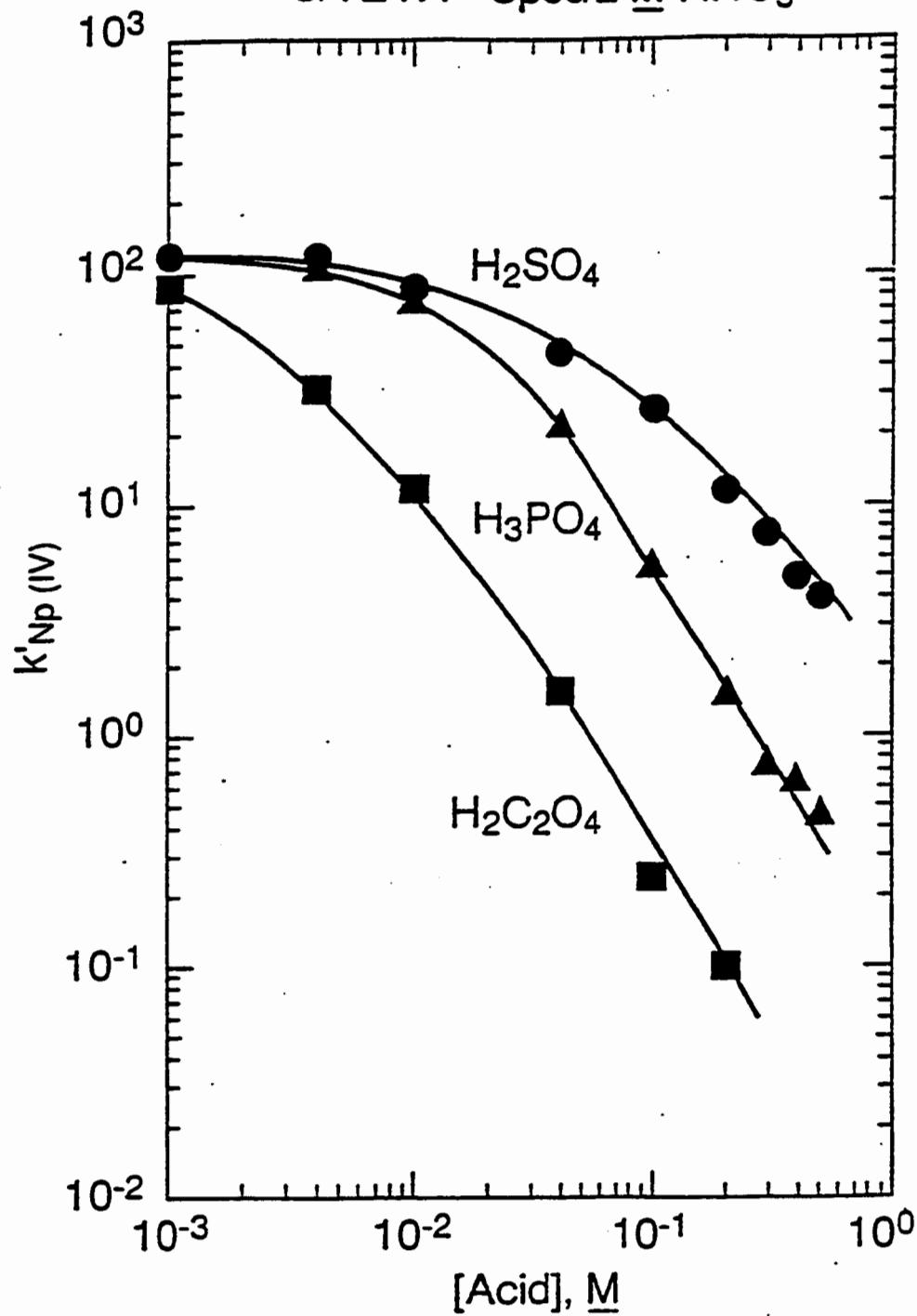
<sup>d</sup> By atomic absorption.

<sup>e</sup> Radiometric.

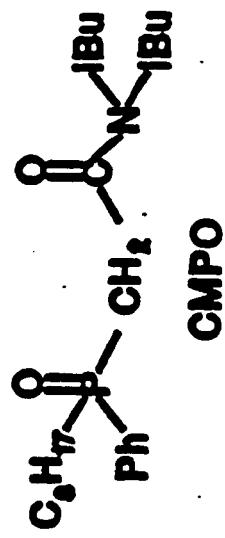
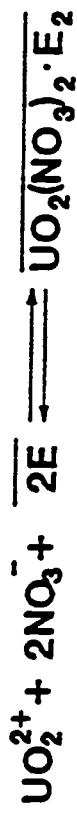
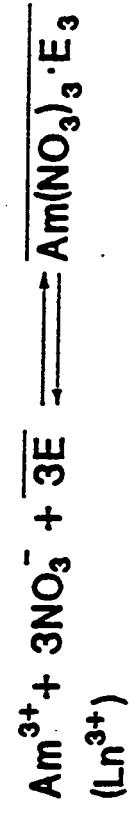
Effect of Matrix Constituents on Uranium Retention  
U/TEVA • Spec/2 M HNO<sub>3</sub>



Effect of Matrix Constituents on Neptunium Retention  
U/TEVA • Spec/2 M HNO<sub>3</sub>



TRU-Spec

**Extractant:****Diluent:** TBP**Extraction Equilibrium:**

**Characteristics of the TRU-Spec extraction chromatographic material and packed columns.**

**Bulk Material**

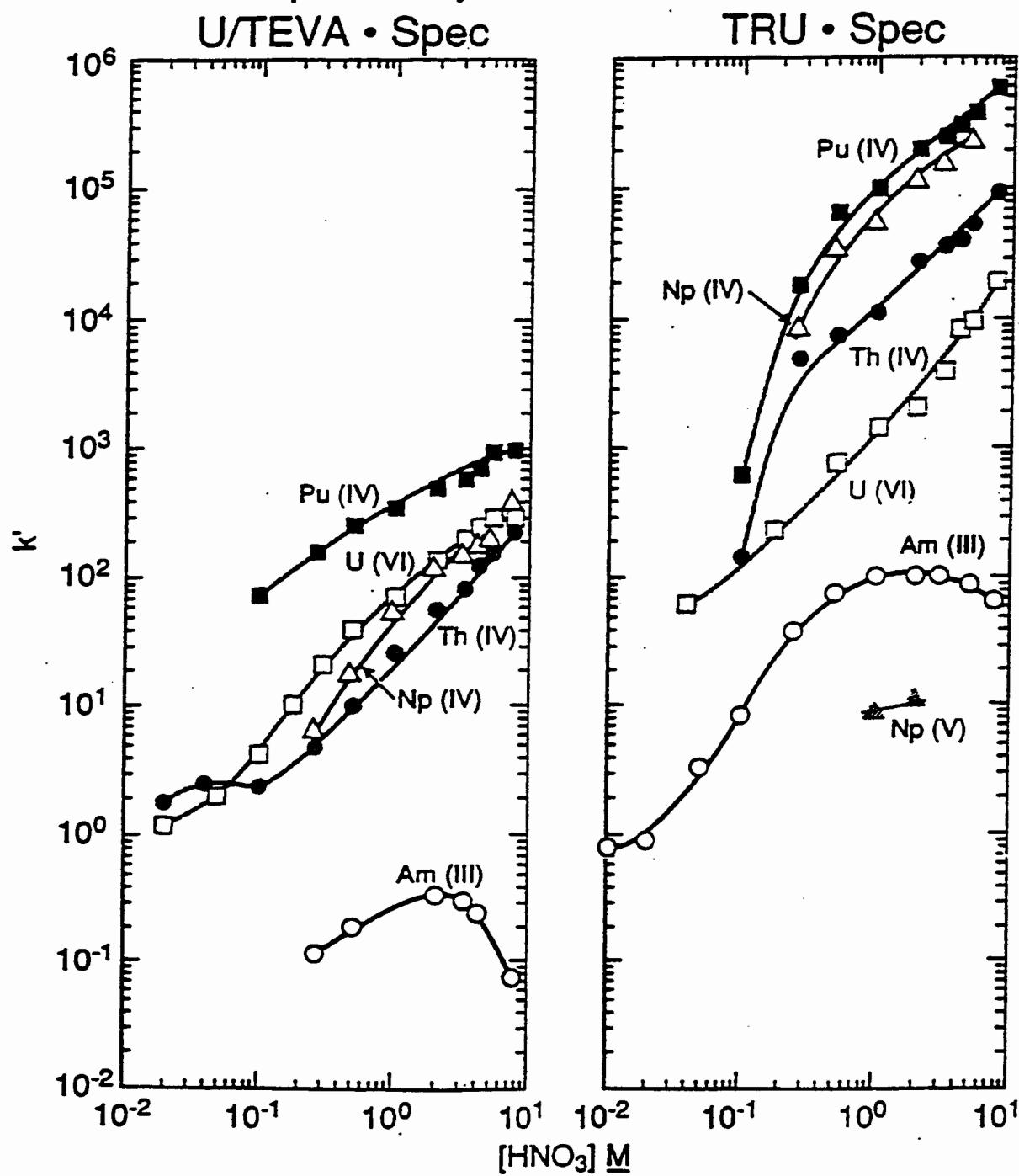
Stationary Phase	0.75 M CMPO in TBP ( $\rho = 0.971$ g/mL)
Support	Amberchrom™ CG-71
Particle Diameter	50-100 $\mu\text{m}$ and 80-160 $\mu\text{m}$
Extractant Loading	40%
Average Density of Extractant-Loaded Beads <sup>a</sup>	1.12 g/mL

**Packed Columns**

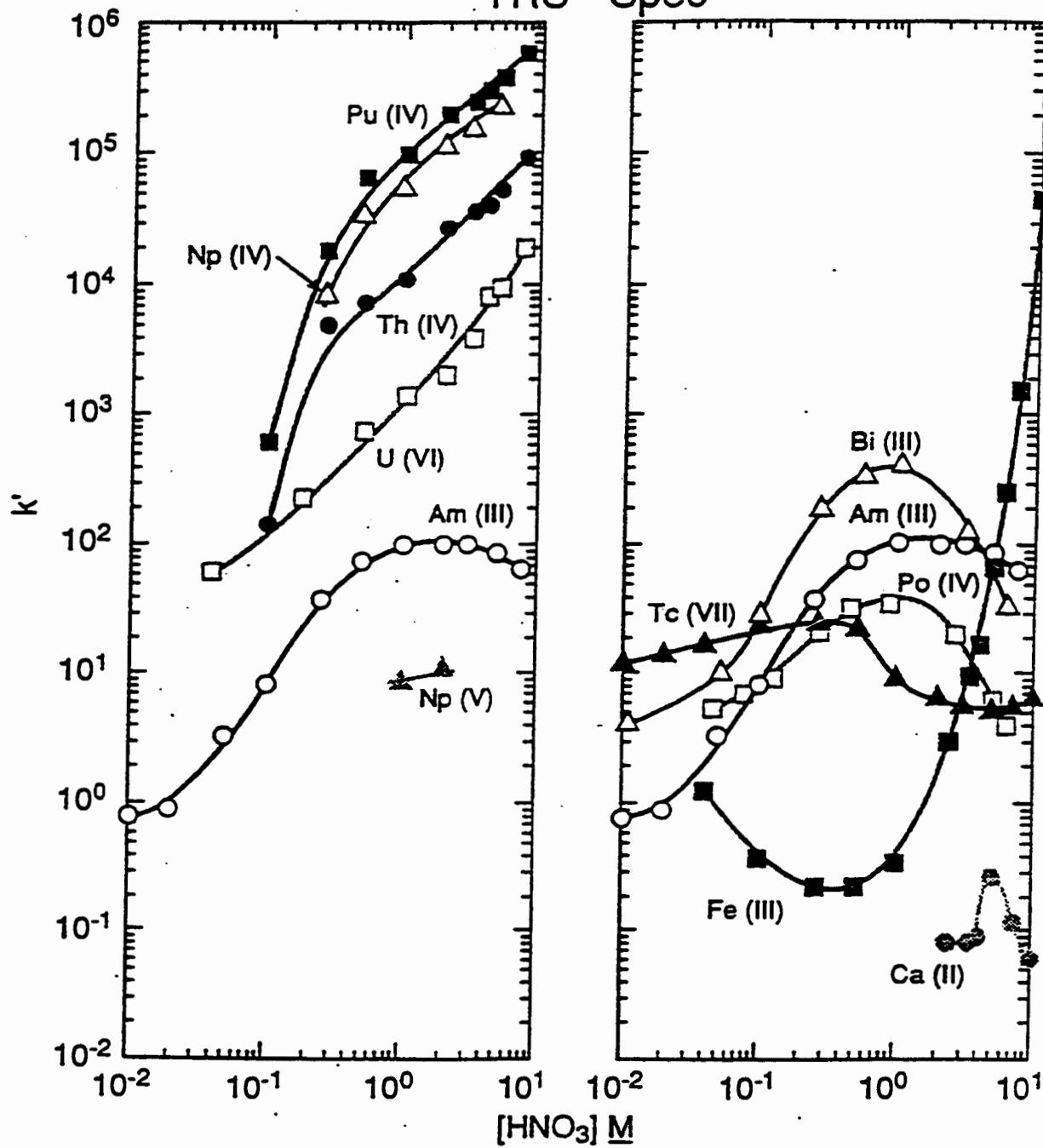
$v_s$	0.152 mL/mL of bed
Bed Density	0.370 g/mL
$v_m$ (also FCV)	0.68 mL/mL of bed
$v_s/v_m$	0.223
Calculated Capacity	5.49 mg Nd or 9.18 mg $^{241}\text{Am}/\text{mL}$ of bed
Experimentally Measured Capacity	4.1 mg Nd or 6.8 mg $^{241}\text{Am}/\text{mL}$ of bed

<sup>a</sup> Picnometric density and flotation density were 1.081 (in water) and 1.158 (in 4.9 M HNO<sub>3</sub>) g/mL, respectively, using the 50-100  $\mu\text{m}$  particle size resin. The calculated density is 1.094 g/mL assuming 100% pore filling and no swelling.

## Acid dependency of $k'$ for various ions at 23°C.



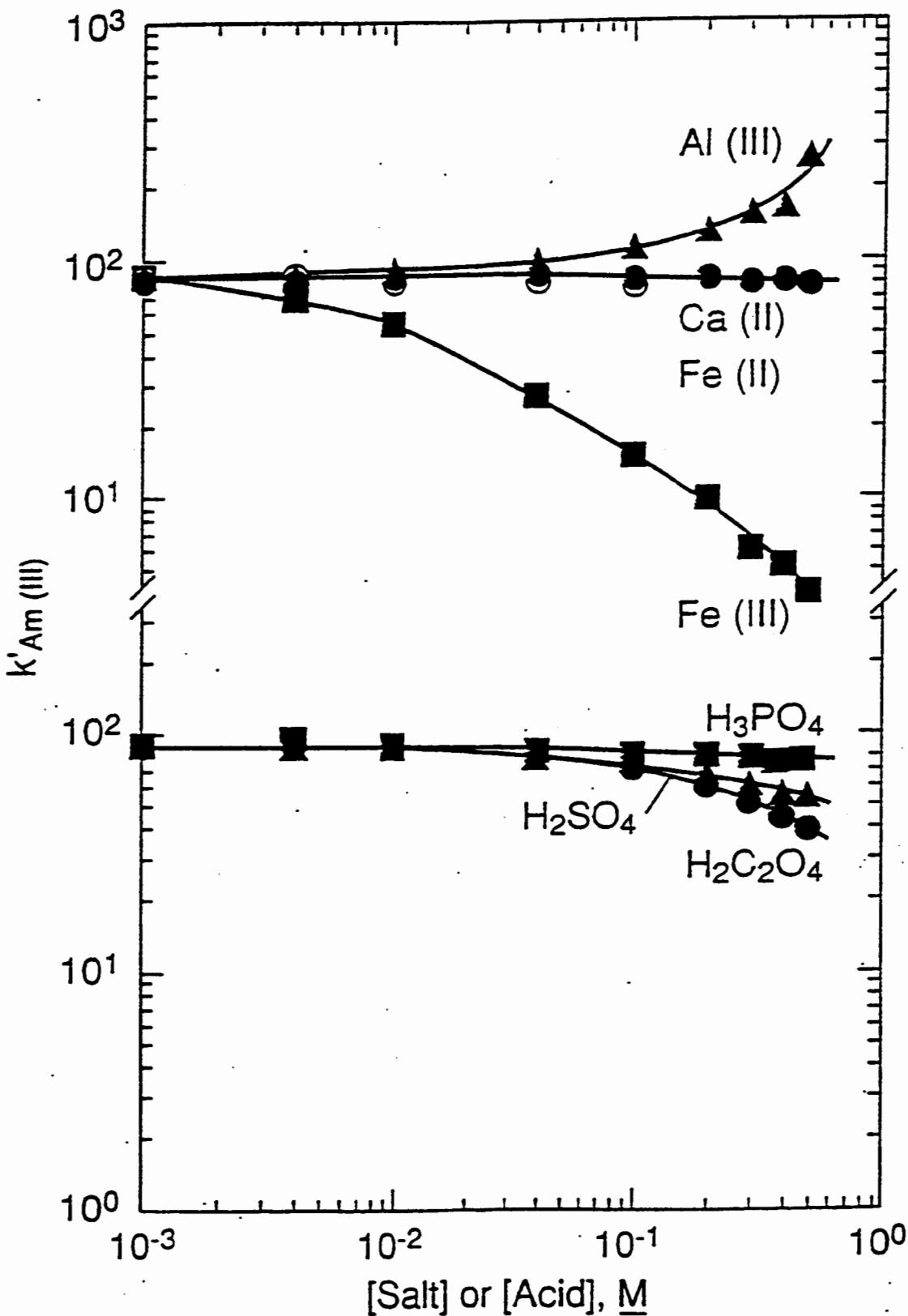
Acid dependency of  $k'$  for various ions at 23-25°C.  
TRU • Spec



21

### Effect of Matrix Constituents on Americium Retention

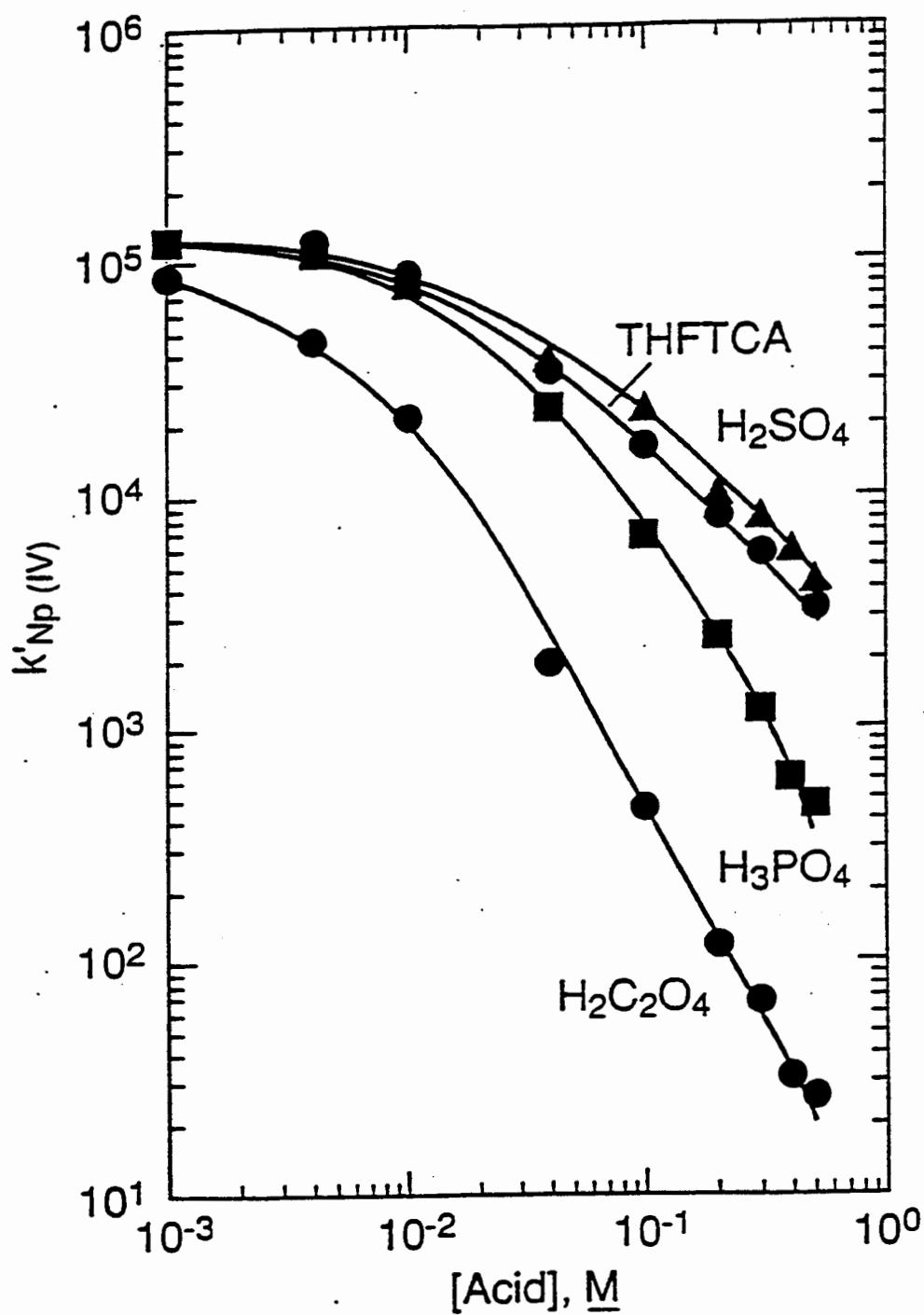
TRU • Spec/2 M HNO<sub>3</sub>



**Effect of Fe on Am Uptake by TRU Spec  
(Reference  $K'_{Am}$  at 2M  $HNO_3$ , 23 °C = 98)**

<u>mg of Fe/10mL</u>	<u>M of Ascorbic Acid</u>	<u><math>K'_{Am}</math></u>
25	-	28
50	-	17
100	-	9.1
25	0.3	88
50	0.3	72
100	0.3	65

Effect of Matrix Constituents on Neptunium Retention  
TRU • Spec/2 M HNO<sub>3</sub>



**Elution of Selected Elements on a TRU-Spec Column  
*(fine particles)***

<u>Element</u>	<u>Fraction Eluting (%) by number of free column volumes</u>						
	1-5	6-10	11-15	16-20	21-25	26-30	31-40
Li	98.4	<19	-	-	-	-	-
Na	92.8	<1.2	-	-	-	-	-
Mg	100	-	-	-	-	-	-
Al	99.8	<2.9	-	-	-	-	-
K	81.8	40.9	-	-	-	-	-
Ca	100	-	-	-	-	-	-
Cr	100	-	-	-	-	-	-
Mn	100	-	-	-	-	-	-
Fe	102	12.3	-	-	-	-	-
Co	100	-	-	-	-	-	-
Ni	100	-	-	-	-	-	-
Cu	100	-	-	-	-	-	-
Sn	100	-	-	-	-	-	-
Sr	100	-	-	-	-	-	-
Y	23.4	76.8	3.5	-	-	-	-
Zr	-	-	-	-	-	-	75.0
Ru	82.6	<19.2	-	-	-	-	-
Rh	100	-	-	-	-	-	-
Ag	100	-	-	-	-	-	-
Cd	100	-	-	-	-	-	-
Ba	100	-	-	-	-	-	-
La	-	-	-	-	-	30.0	72.0
Ce	-	-	-	-	-	>25.0	75.0
Pr	-	-	-	-	-	-	100
Nd	-	-	-	-	-	-	96.0
Sm	-	-	-	-	-	-	100
Eu	-	-	-	-	-	-	>99
Hg	(100)	(60)	(19)	-	-	-	-
Pb	100	-	-	-	-	-	-
Am*	-	-	-	-	-	-	>99

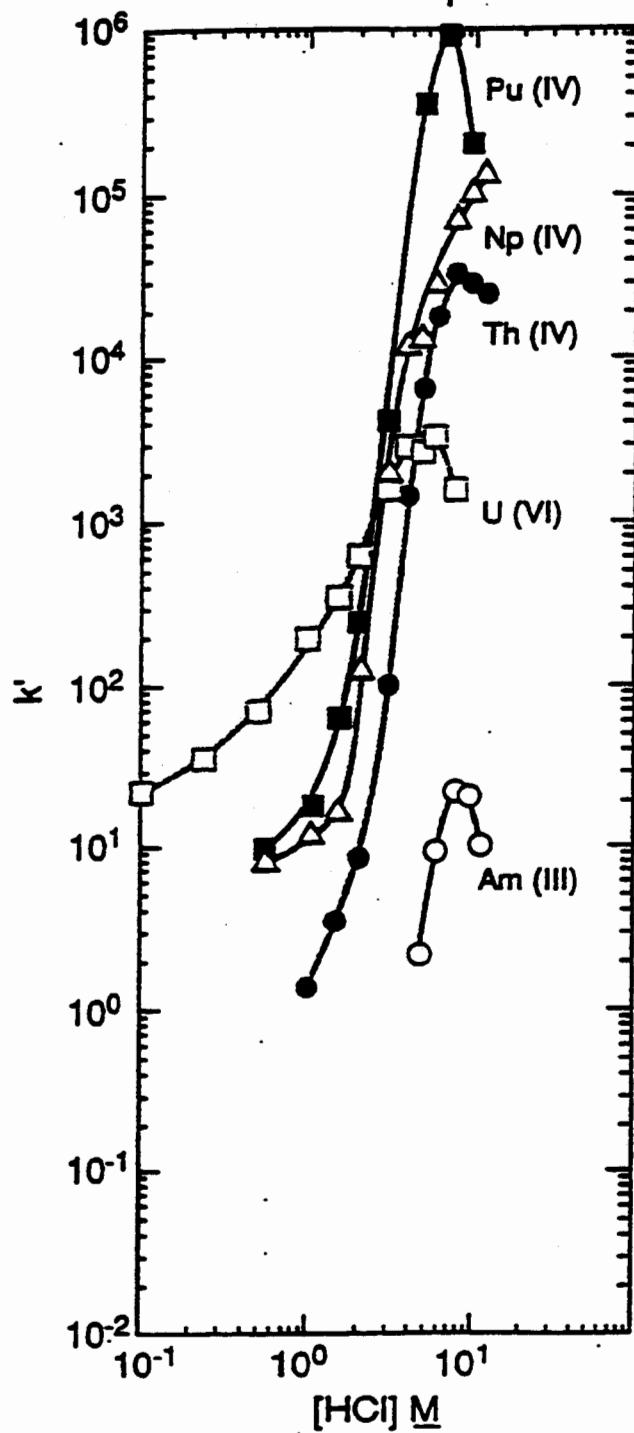
\* radiometric

1-30 f.c.v. : 2M HNO<sub>3</sub>

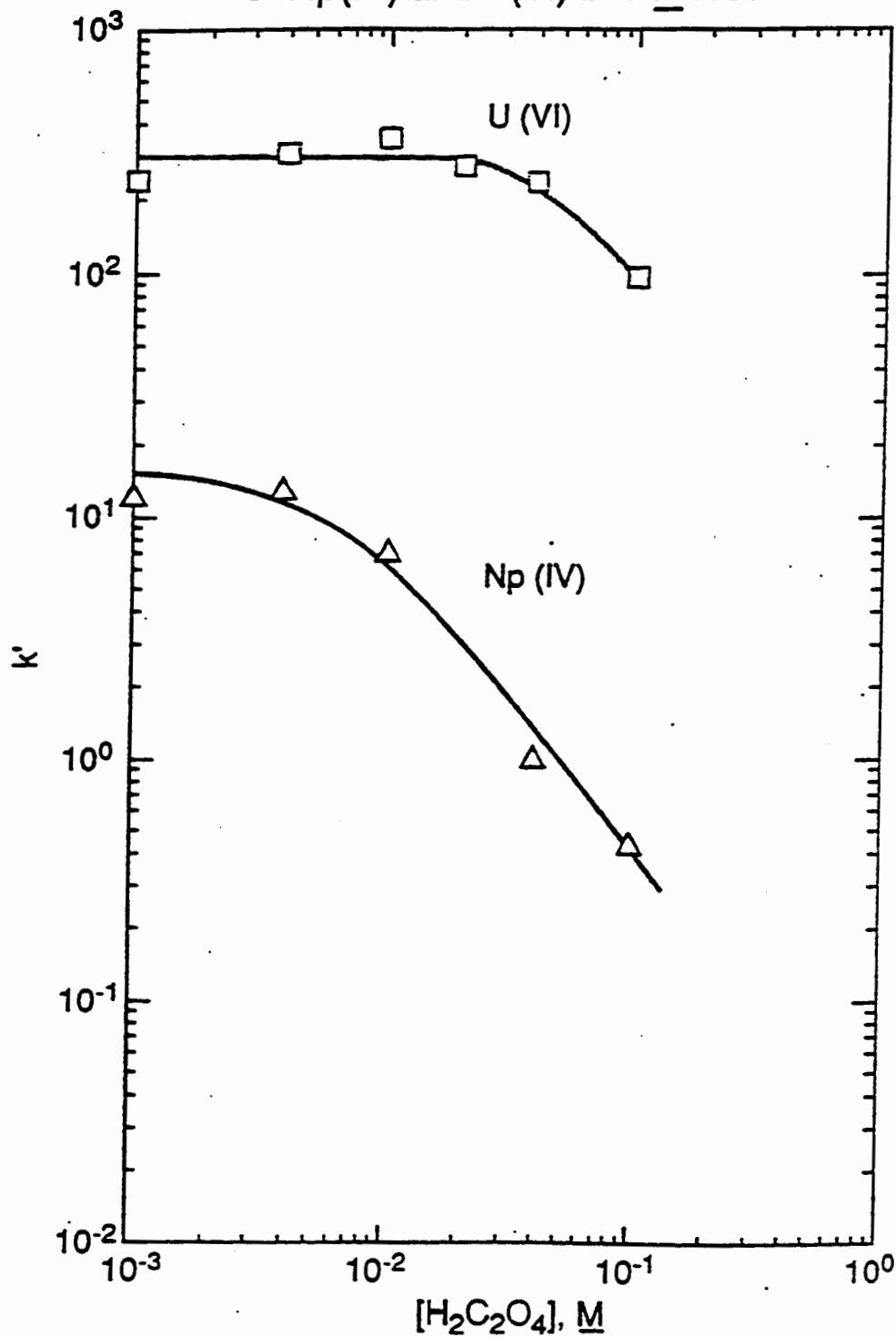
31-40 f.c.v. : 0.05M HNO<sub>3</sub>

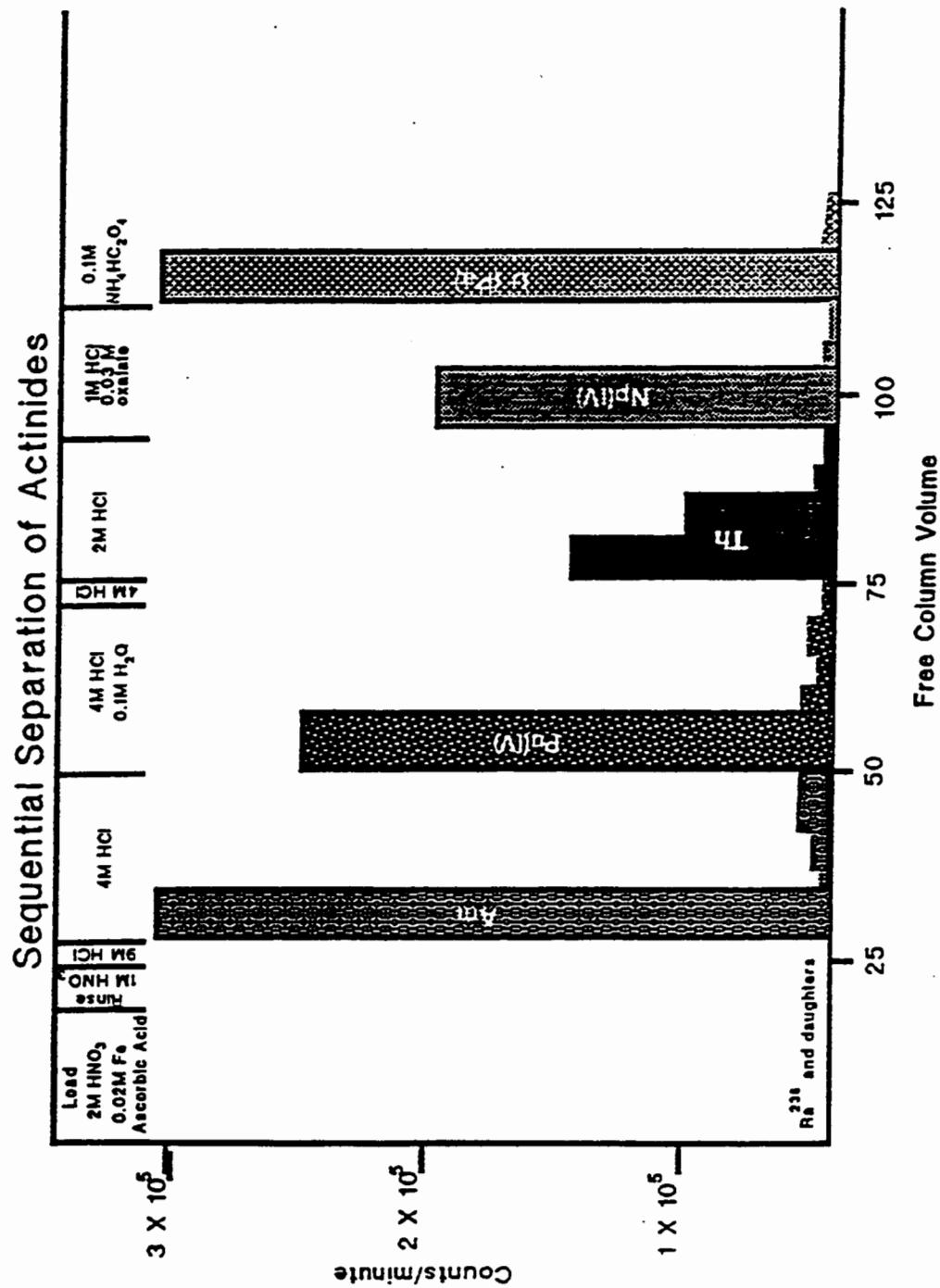
**note:** Because of uncertainties inherent in the ICP-AES method, the fractions shown for each element may not total to 100%. Values in parentheses are subject to considerable uncertainty and are intended only as a rough guide.

Acid dependency of  $k'$   
for various ions at 23°C.  
TRU • Spec



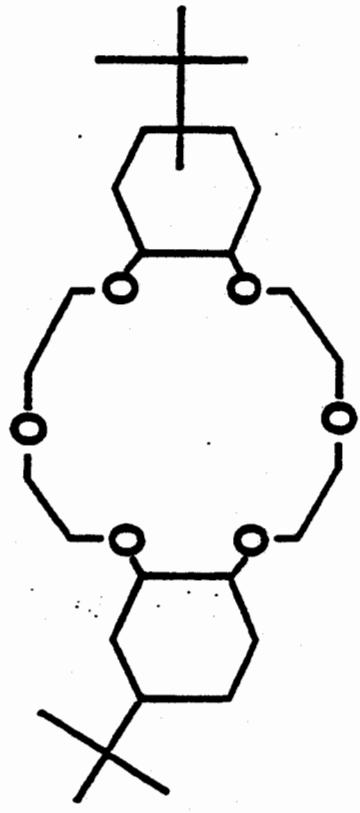
TRU • Spec/HCl  
Effects of oxalic acid on  $k'$   
of Np(IV) and U(VI) in 1 M HCl





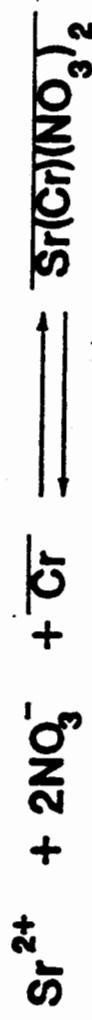
Sr·Spec

Extractant: bis-*t*-Butyl-*cis*-Dicyclohexano-18-Crown-6



Diluent: 1-Octanol ( $C_8H_{17}OH$ )

Extraction Equilibrium:



**CHARACTERISTICS OF STRONTIUM-SELECTIVE EXTRACTION  
CHROMATOGRAPHIC MATERIAL AND PACKED COLUMNS**

BULK MATERIAL	
Stationary Phase	1.0 M DtBuCH18C6 in 1-octanol <sup>a</sup> ( $\rho = 0.912 \text{ g/mL}$ )
Support	Amberchrom™ CG-71 or Amberlite™ XAD-7
Particle Sizes (diam.)	50-100 $\mu\text{m}$ ; 80-160 $\mu\text{m}$ ; 100-125 $\mu\text{m}$
Extractant Loading	40 weight percent
Average Density of Extractant-Loaded Beads <sup>b</sup>	1.12 g/ml

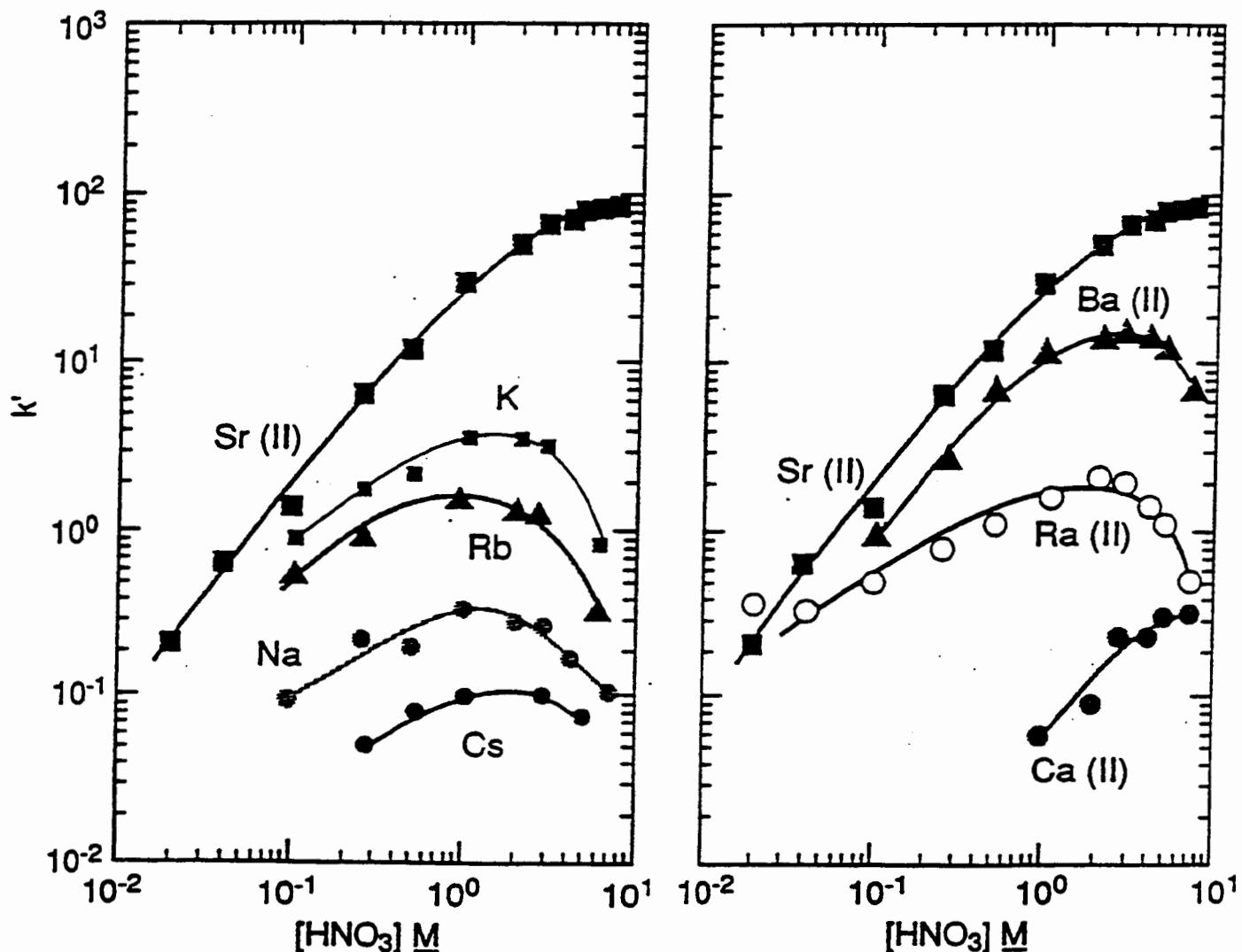
PACKED COLUMN		Particle Size
	50-100 $\mu\text{m}$	80-160 $\mu\text{m}$
$v_s$ , ml/ml of Bed <sup>c</sup>	$0.146 \pm 0.004$	$0.152 \pm 0.004$
Bed Density (g/ml)	0.33	0.35
$v_m$ (Also F.C.V.), ml/ml of Bed <sup>c</sup>	$0.71 \pm 0.01$	$0.69 \pm 0.01$
$v_s/v_m$ <sup>c</sup>	$0.21 \pm 0.01$	$0.22 \pm 0.01$
<b>Capacity:</b>		
Calc. mg Sr/ml of Bed	12.8	13.3
Exp. mg Sr/ml of Bed	10.6	—

<sup>a</sup> 1 M concentration of DtBuCH18C6 in 1-octanol is close to a saturated solution; therefore, the Sr retention from nitric acid is close to the maximum that can be achieved with this system.

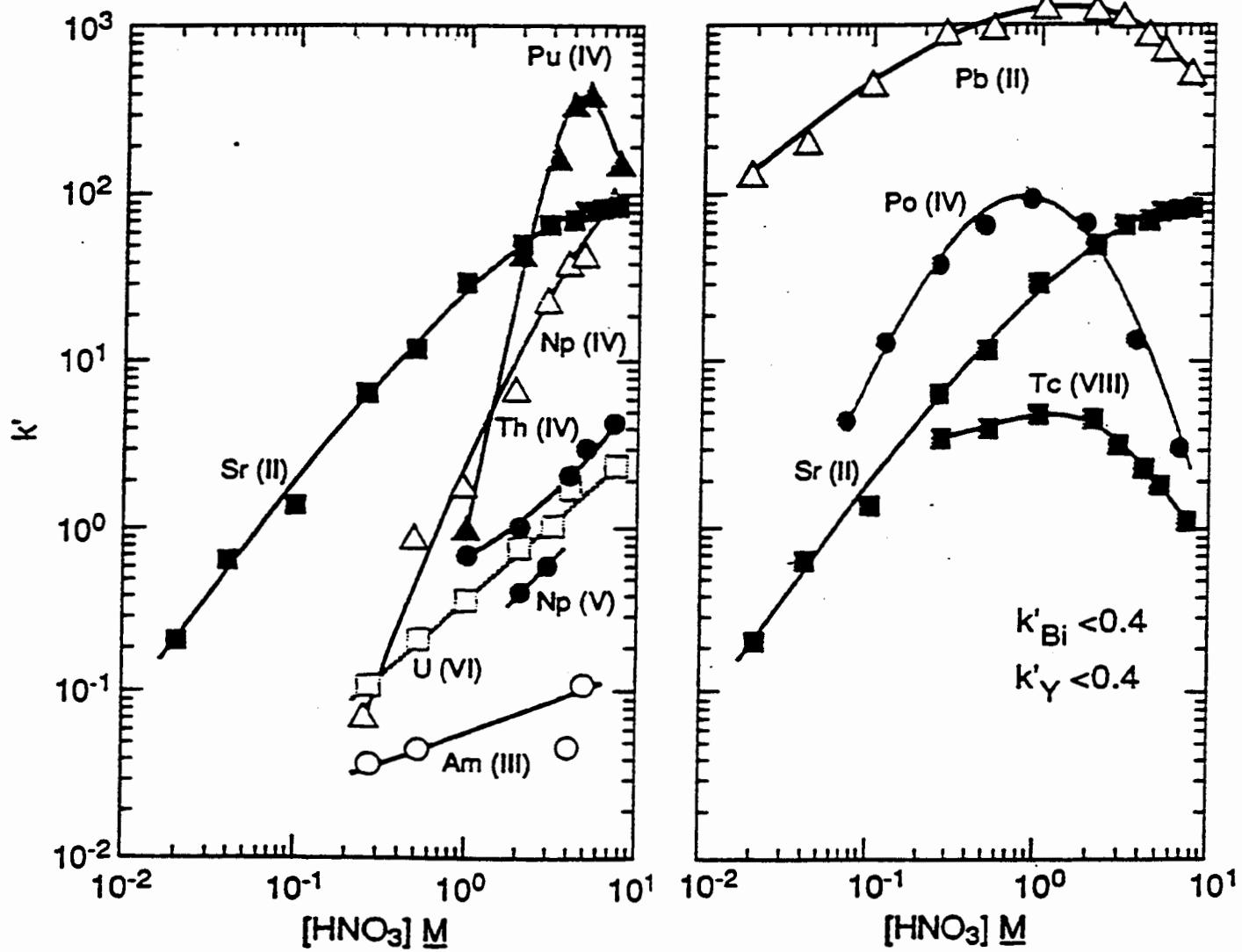
<sup>b</sup> Picnometric density and flotation density values were 1.079 and 1.157 g/ml, respectively, using the 50-100  $\mu\text{m}$  particle size resin. The calculated density is 1.094 g/ml, assuming 100% pore filling and no swelling.

<sup>c</sup> Average of five measurements: error is one  $\sigma$ .

Acid dependency of  $k'$  for various ions at 23-25°C.  
 Sr • Spec

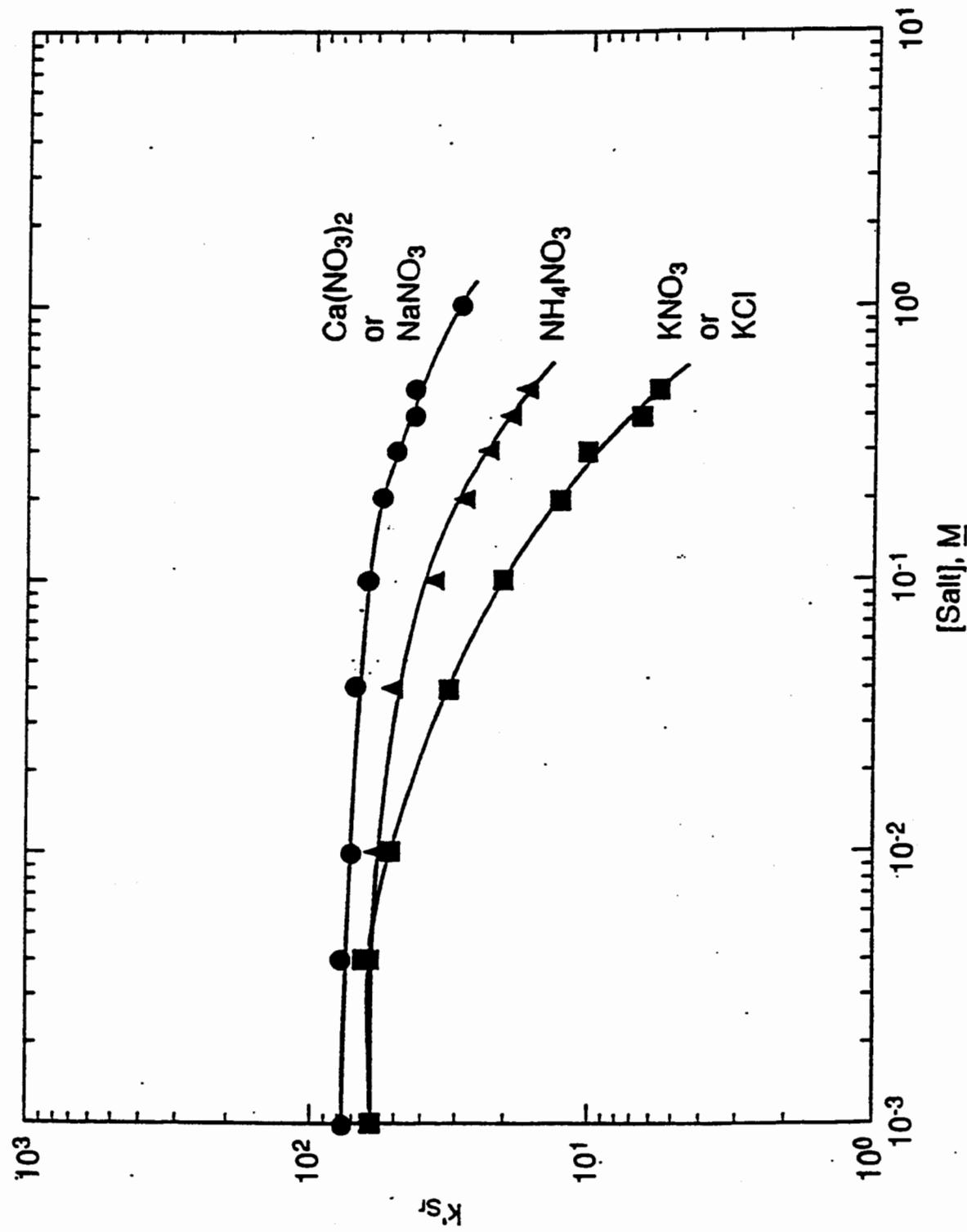


Acid dependency of  $k'$  for various ions at 23-25°C.  
 Sr • Spec

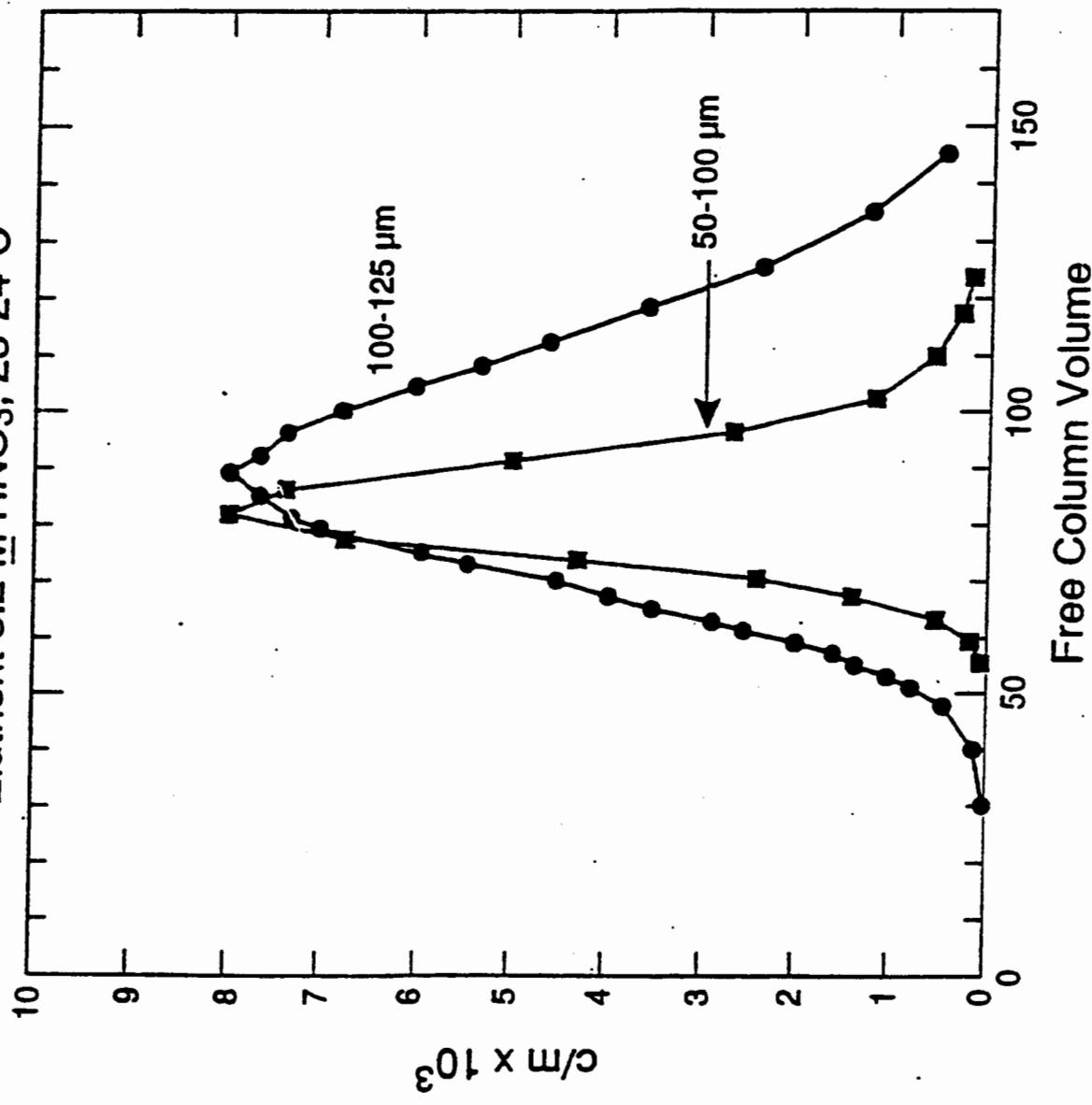


Effect of Matrix Constituents on Strontium Retention  
Sr • Spec/3 M HNO<sub>3</sub>

32



Comparison of Elution Curves for  $\text{Sr}^{2+}$   
for Two Particles of Sr-Spec  
Elutriant 3.2 M  $\text{HNO}_3$ , 23-24°C



ELUTION BEHAVIOR OF COMMON ELEMENTS AND FISSION  
PRODUCTS ON THE STRONTIUM-SELECTIVE RESIN

PERCENT OF ELEMENT FOUND IN F.C.V. #

Element	3 M HNO <sub>3</sub> - 0.01 M Oxalic Acid						0.05 M HNO <sub>3</sub>	
	1-5	6-10	11-15	16-20	21-25	26-30	31-40	F.C.V.
Li	100	—	—	—	—	—	—	—
Na	100	—	—	—	—	—	—	—
K	66	35	—	—	—	—	—	—
Rb	100	—	—	—	—	—	—	—
Cs	100	—	—	—	—	—	—	—
Mg	100	—	—	—	—	—	—	—
Ca	100	—	—	—	—	—	—	—
Sr	—	—	—	—	—	—	—	99
Ba	—	—	53	42	6	0.7	—	—
Ra	—	99	—	1	—	—	—	—
Al	100	—	—	—	—	—	—	—
Cr	100	—	—	—	—	—	—	—
Mn	100	—	—	—	—	—	—	—
Fe	99	0.6	0.2	0.4	—	—	—	—
Co	100	—	—	—	—	—	—	—
Ni	100	—	—	—	—	—	—	—
Cu	100	0.2	—	—	—	—	—	—
Zn	100	0.2	—	—	—	—	—	—
Y	100	0.1	—	—	—	—	—	—
Zr	91	0.4	0.2	—	—	—	—	—
Mo	—	84	—	16	—	—	—	—
Tc	57	43	—	—	—	—	—	—
Ru	100	—	—	—	—	—	—	—
Rh	100	—	—	—	—	—	—	—
Pd	100	—	—	—	—	—	—	—
Ag	15	88	2	—	—	—	—	—
Cd	100	0.1	—	—	—	—	—	—
La-Eu	100	0.1	—	—	—	—	—	—
Hg	5	5	19	40	19	10	5	—

Column parameters: Particle size = 50-100  $\mu\text{m}$ , Bed Volume = 1.0  $\text{cm}^3$ , Bed height = 5.0 cm,  
and 1 F.C.V. = 0.60 mL.