

Removing a Needle from a Haystack with a Small Fork: Or can a series of small columns handle a kilogram of matrix?

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Recovery of ^{225}Ac from Aged Light Water Breeder Reactor Fuel

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-A chemical process has been developed and patented, no. 7,157,061, to separate ^{225}Ac from dissolved LWBR fuel (97% ^{232}Th and 3% ^{233}U , as oxides)

-Laboratory scale studies funded by CRADA with Northstar Nuclear Medicine completed on actual aged LWBR fuel pellets

-Aged ^{233}U , via in-growth of ^{229}Th , is the best source of ^{225}Ac

-LWBR fuel, 97% ^{232}Th , thought unusable for ^{225}Ac recovery because the ^{229}Th is not isolatable.

Decay of ^{233}U

^{233}U 1.59×10^5 years

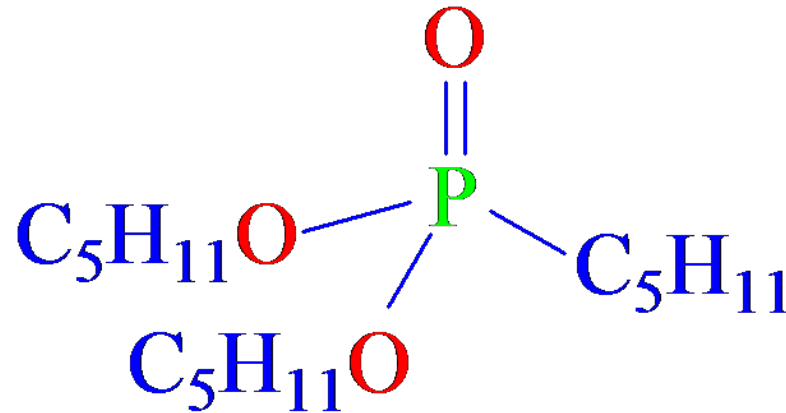


^{229}Th 7.3×10^3 years

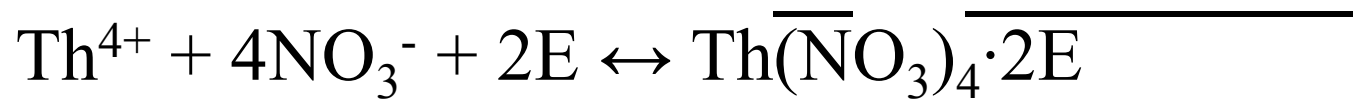


^{225}Ra 14.8 days \longrightarrow ^{225}Ac 10.0 days

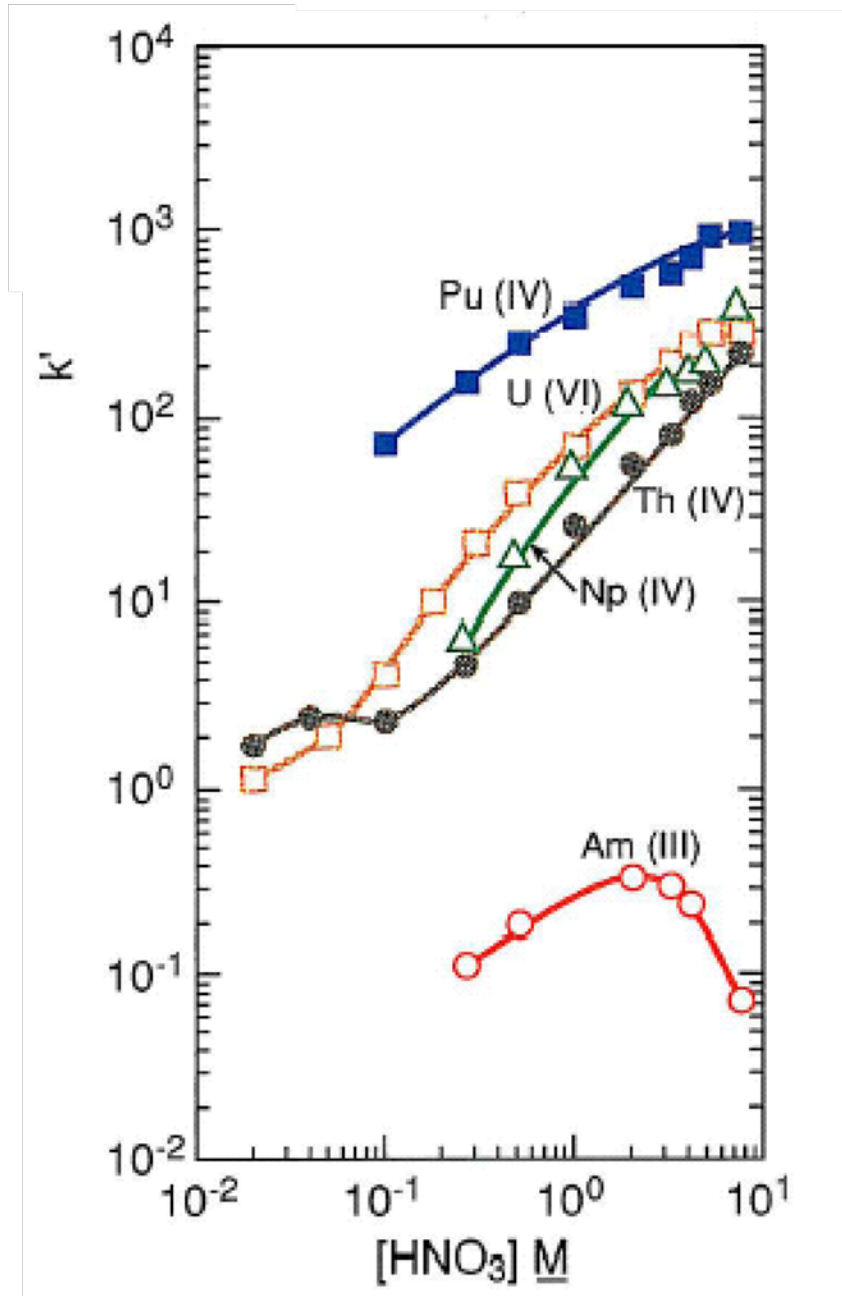
Diamyl amyolphosphonate (DA[AP])



- Very selective for Th and U from acidic nitrate media

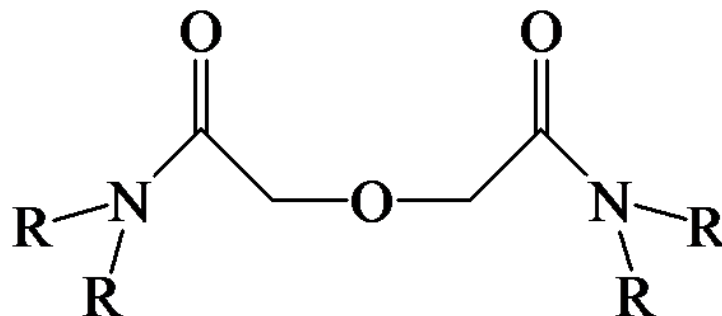


k' for Actinides on UTEVA Resin vs HNO_3



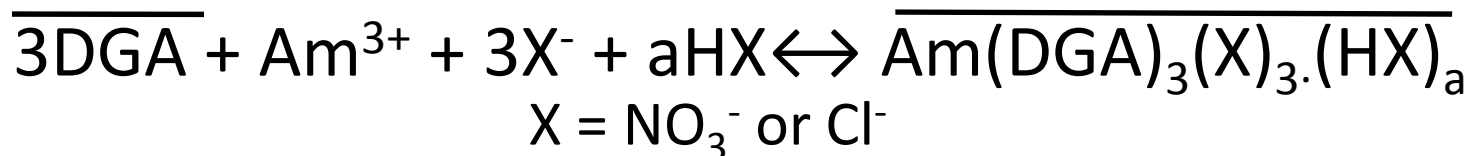
UTEVA Resin is
40% (w:w) DA[AP]
on an inert support

Diglycolamide Extractant



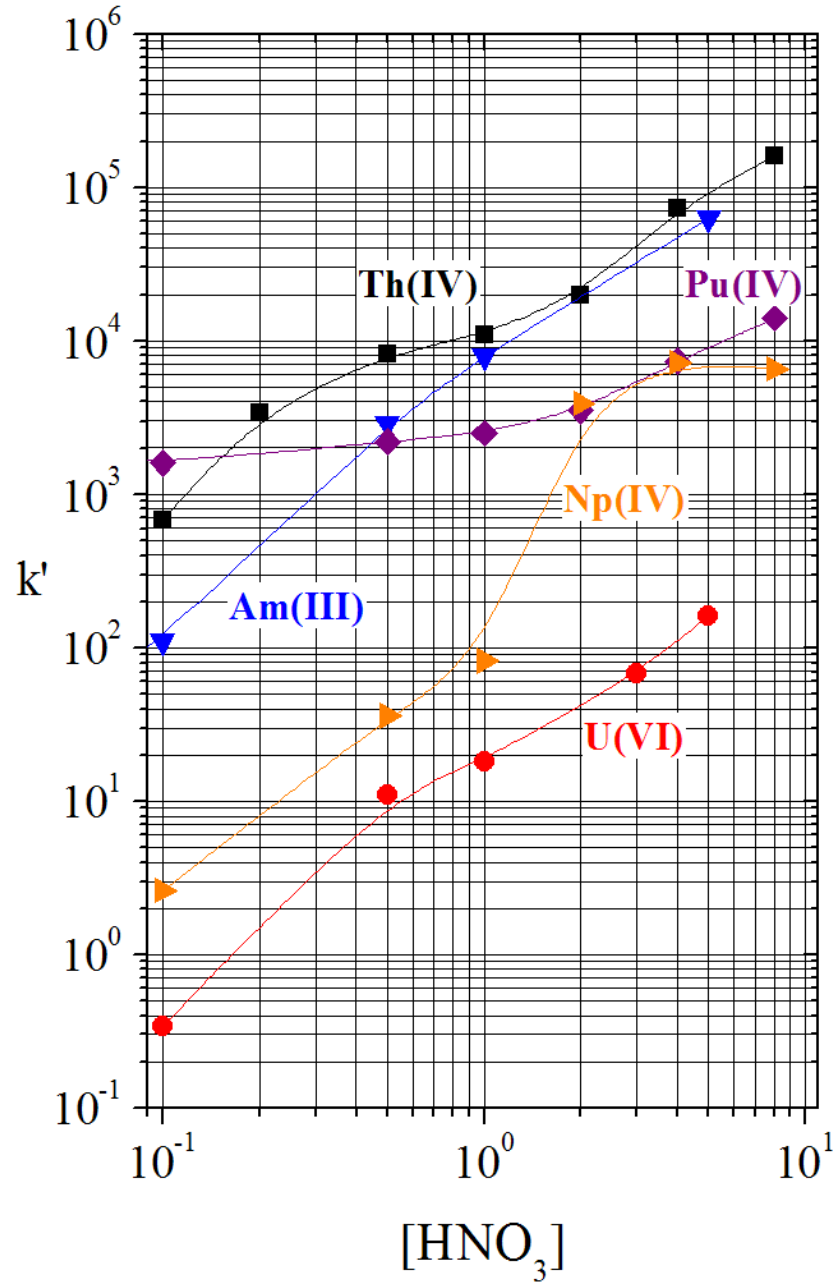
R = *n*-octyl

DGA or DGA, Normal



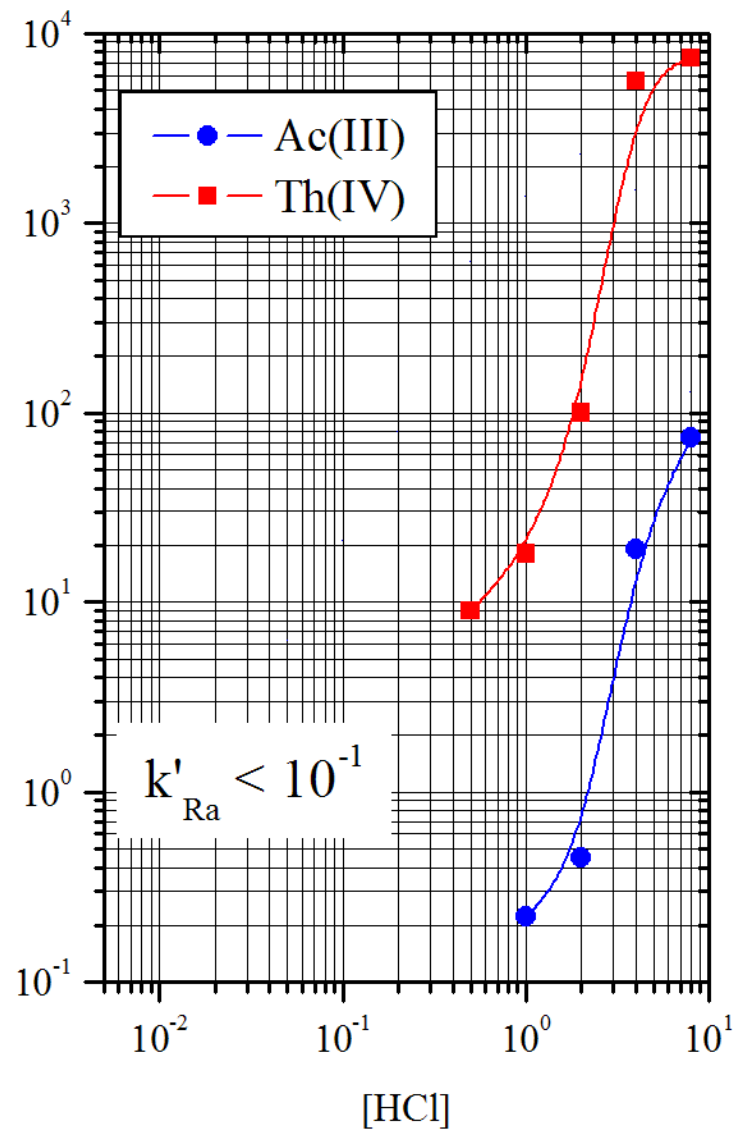
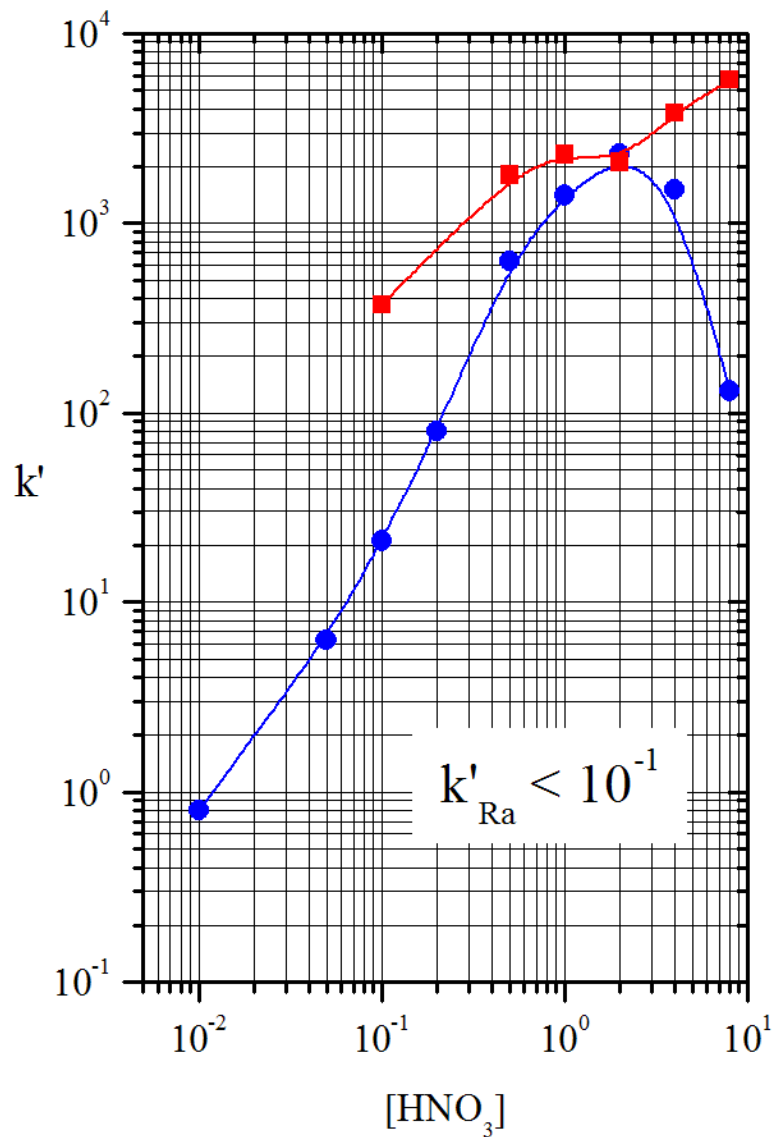
Unique Selectivity: An(III) ~ An(IV) > U(VI)

Unique Selectivity of DGA Resin, Normal

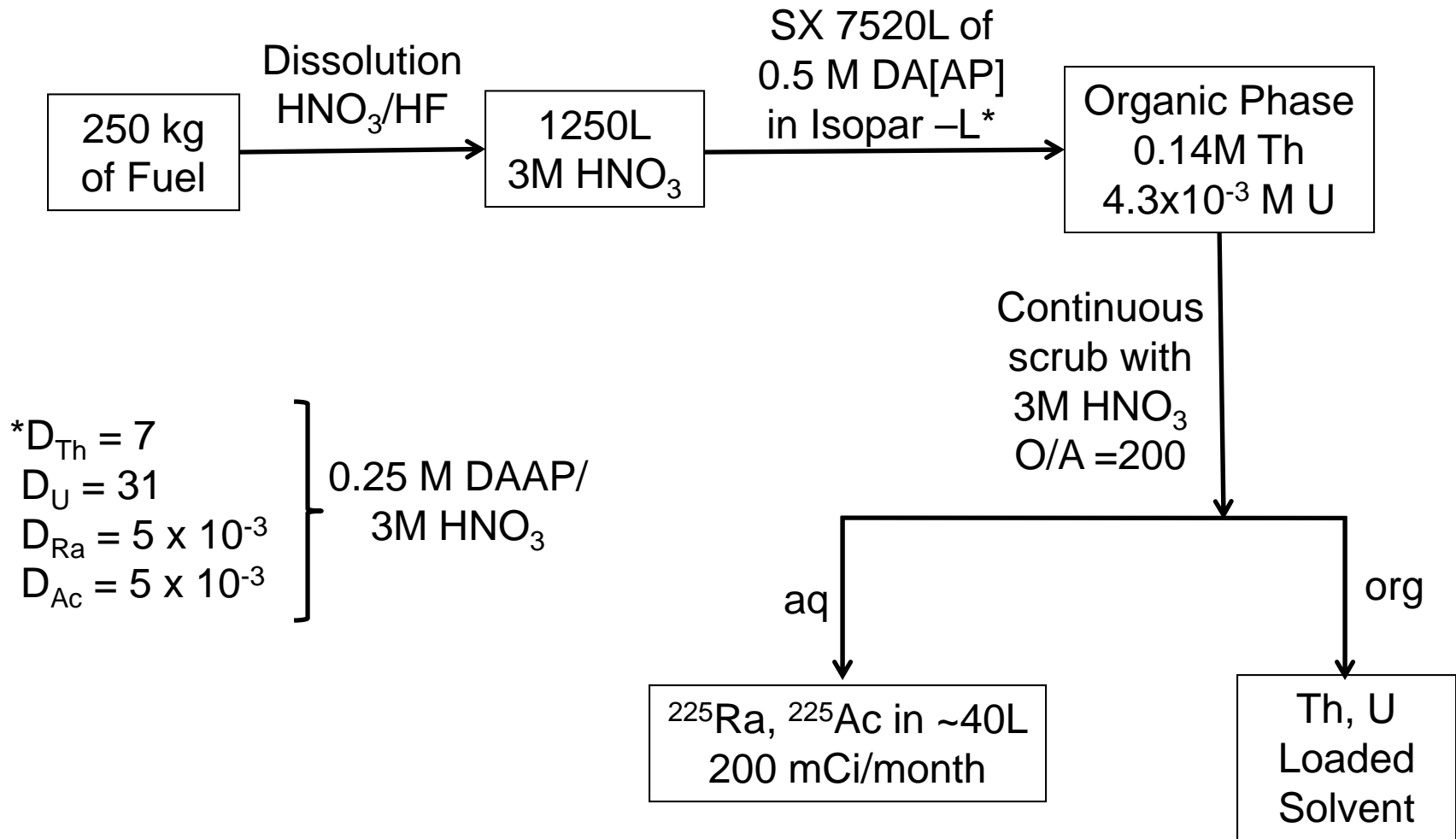


k' Ra(II), Ac(III), and Th(IV) vs. $[\text{HNO}_3]$ or $[\text{HCl}]$ on DGA Resin

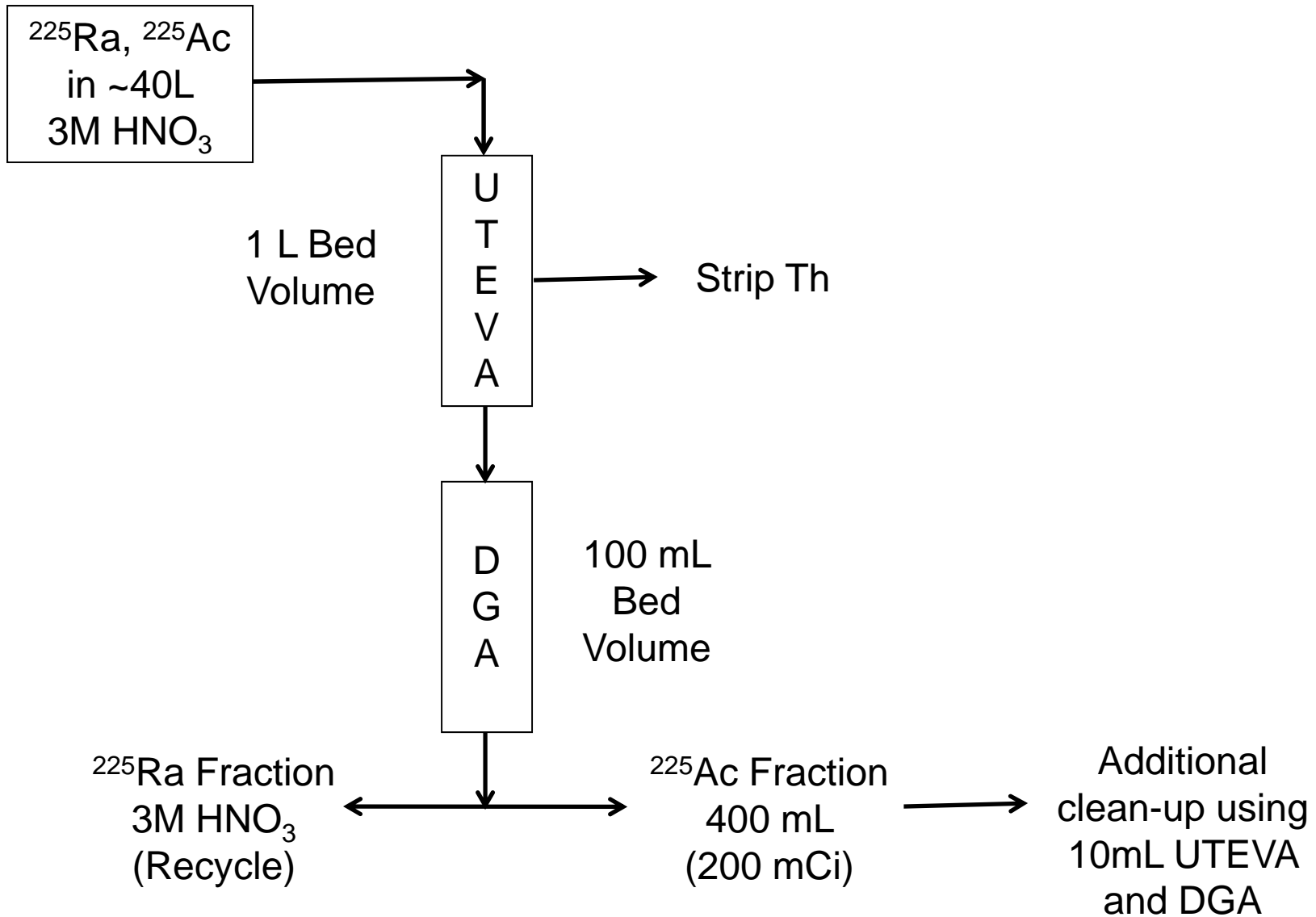
50-100 μm resin, 1 h contact time, 22(2) $^\circ\text{C}$



Conceptual Flowsheet for the Recovery of ^{225}Ra and ^{225}Ac from INL Breeder Reactor Fuel



^{225}Ra and ^{225}Ac Flowsheet (continued)



Flowsheet Tested on 75g of LWBR Fuel

-75 g of fuel was dissolved with 150 mL of 13M HNO₃/0.03M HF in a PARR Reactor at 107°C for 24 hours

-Overall recovery of ²²⁵Ac (80 μCi) 80% yield

-D.F. from ^{229,232}Th and ²³³U > 10¹¹

-D.F. from Ra > 10⁸

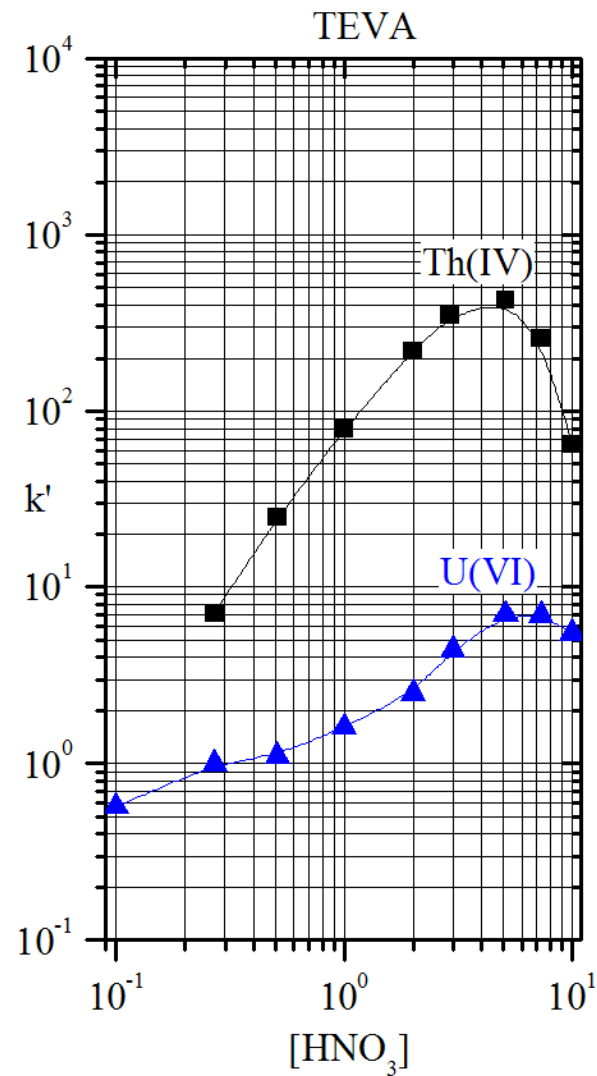
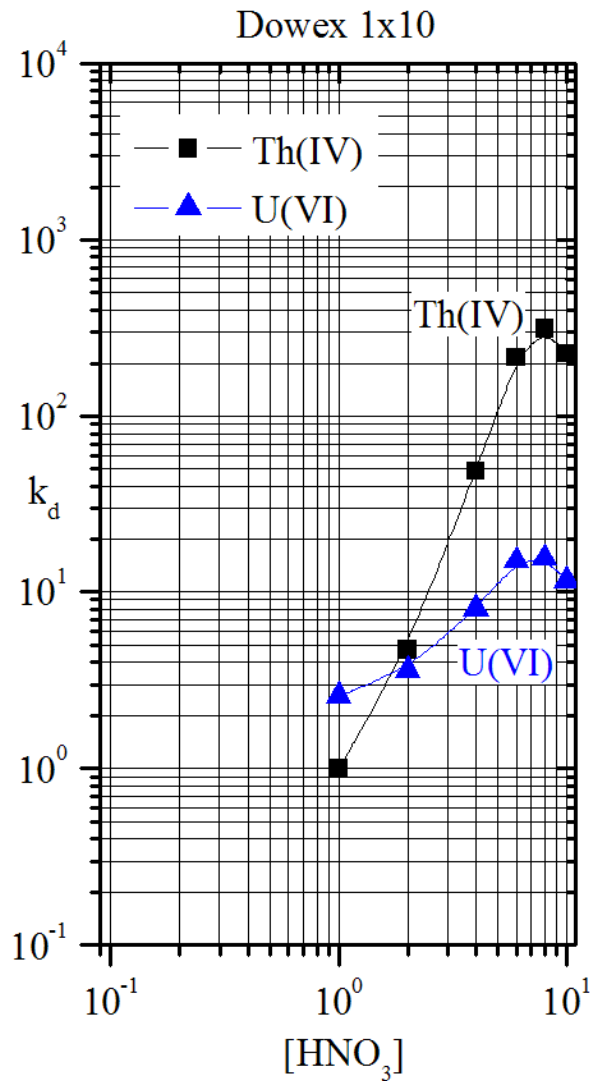
-Impurities <<ug/mCi of ²²⁵Ac

Separation of Trace Th from Macro Uranium

- ^{234}Th from ^{238}U (useful laboratory tracer)

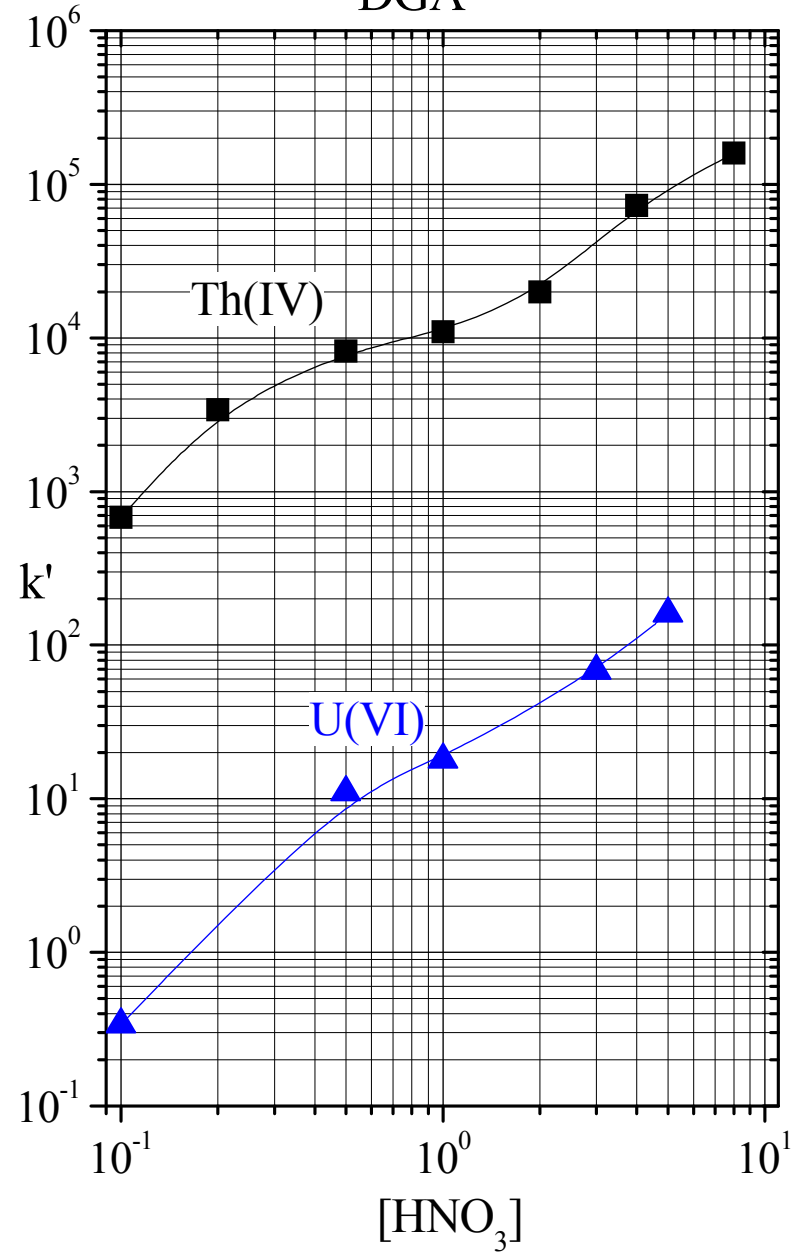
- ^{229}Th from ^{233}U (parent of ^{225}Ac , medical radioisotope)

Th/U Separation Factor

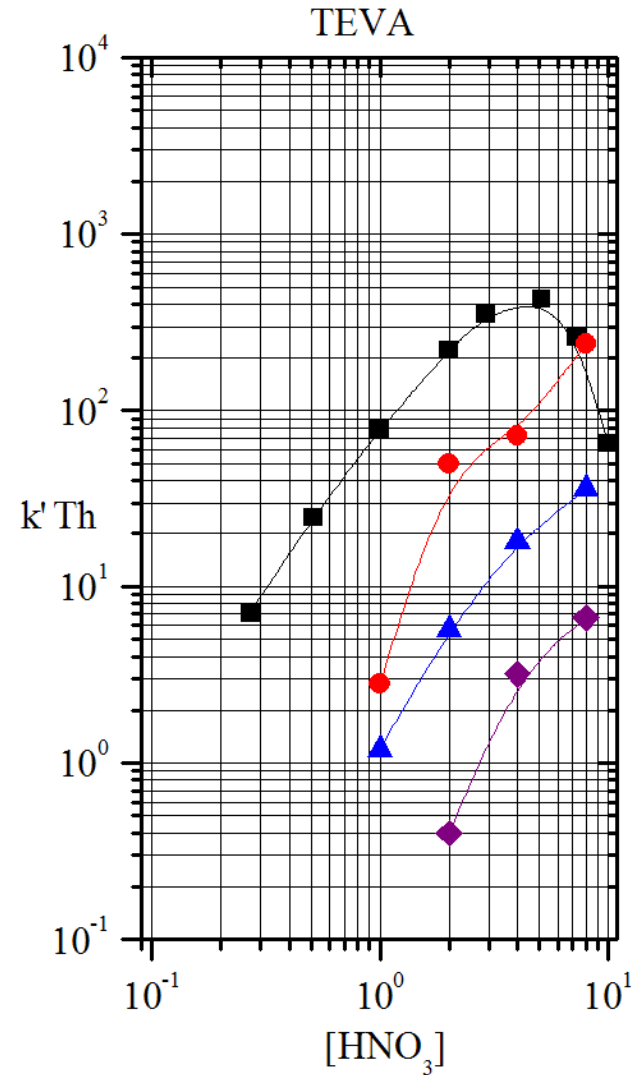
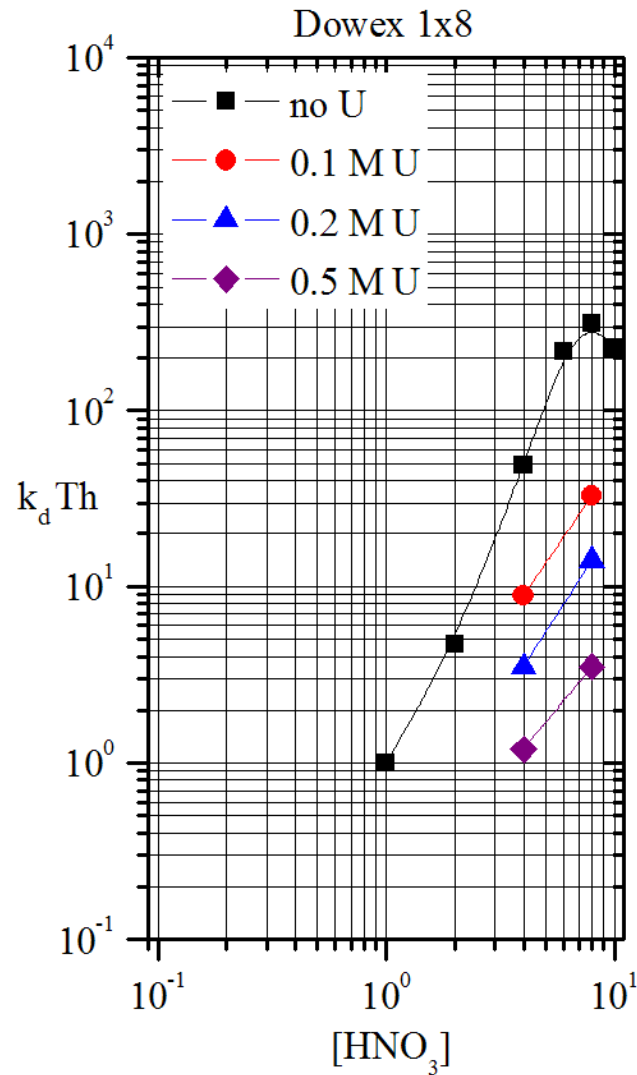


Th/U Separation Factor

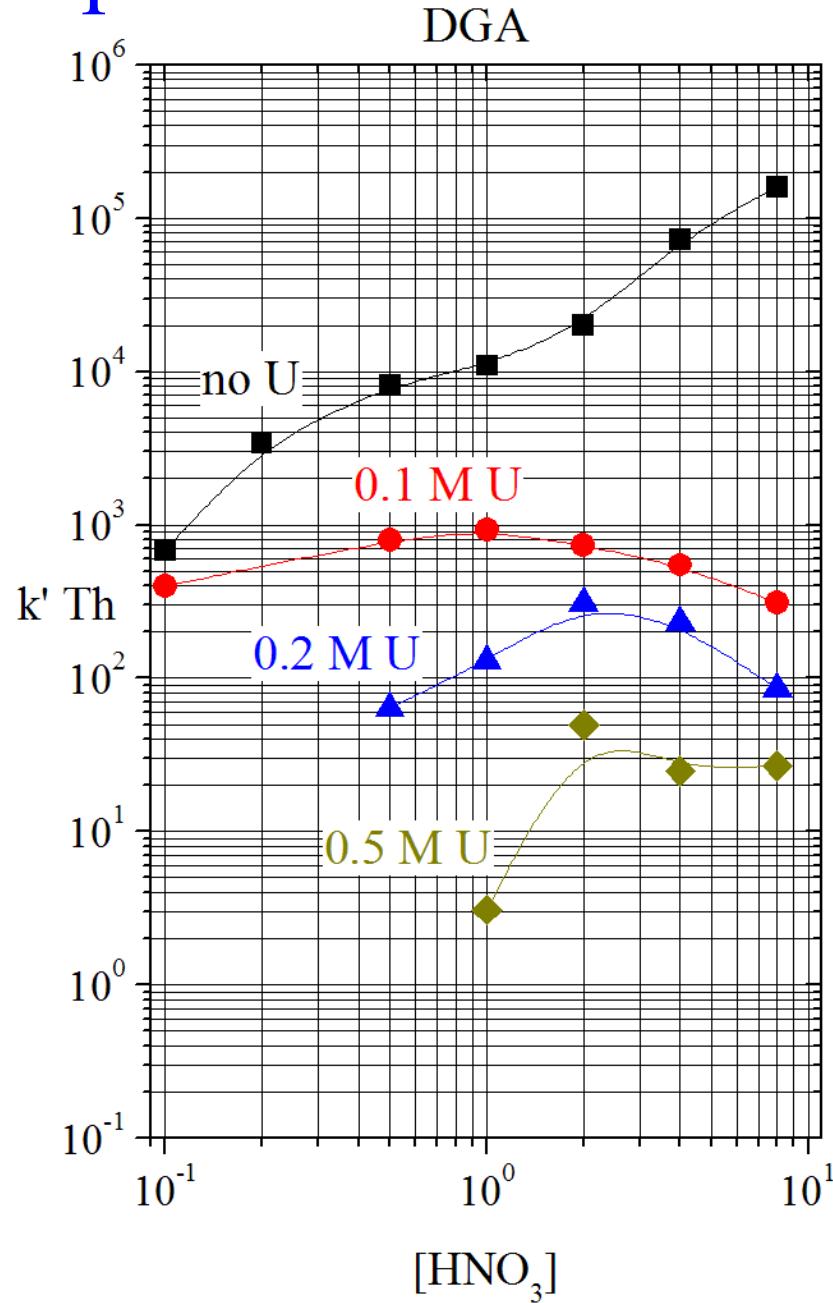
DGA



Th uptake vs U concentration



Th uptake vs U concentration



Optimization for k' for Th

<u>Resin</u>	<u>k'</u>	<u>Acid</u>
Dowex 1x8	300	8M HNO ₃
TEVA	400	4M HNO ₃
DGA	>10 ⁵	8M HNO ₃
	>10 ⁴	4M HNO ₃

Optimization for k' for Th from 0.2M U

<u>Resin</u>	<u>k'</u>	<u>Acid</u>
Dowex 1x8	14	8M HNO ₃
TEVA	20	4M HNO ₃
DGA	300	2M HNO ₃

Decay of ^{238}U

^{238}U 4.47×10^9 years



^{234}Th
24.1 days



^{234}Pa
1.18 min

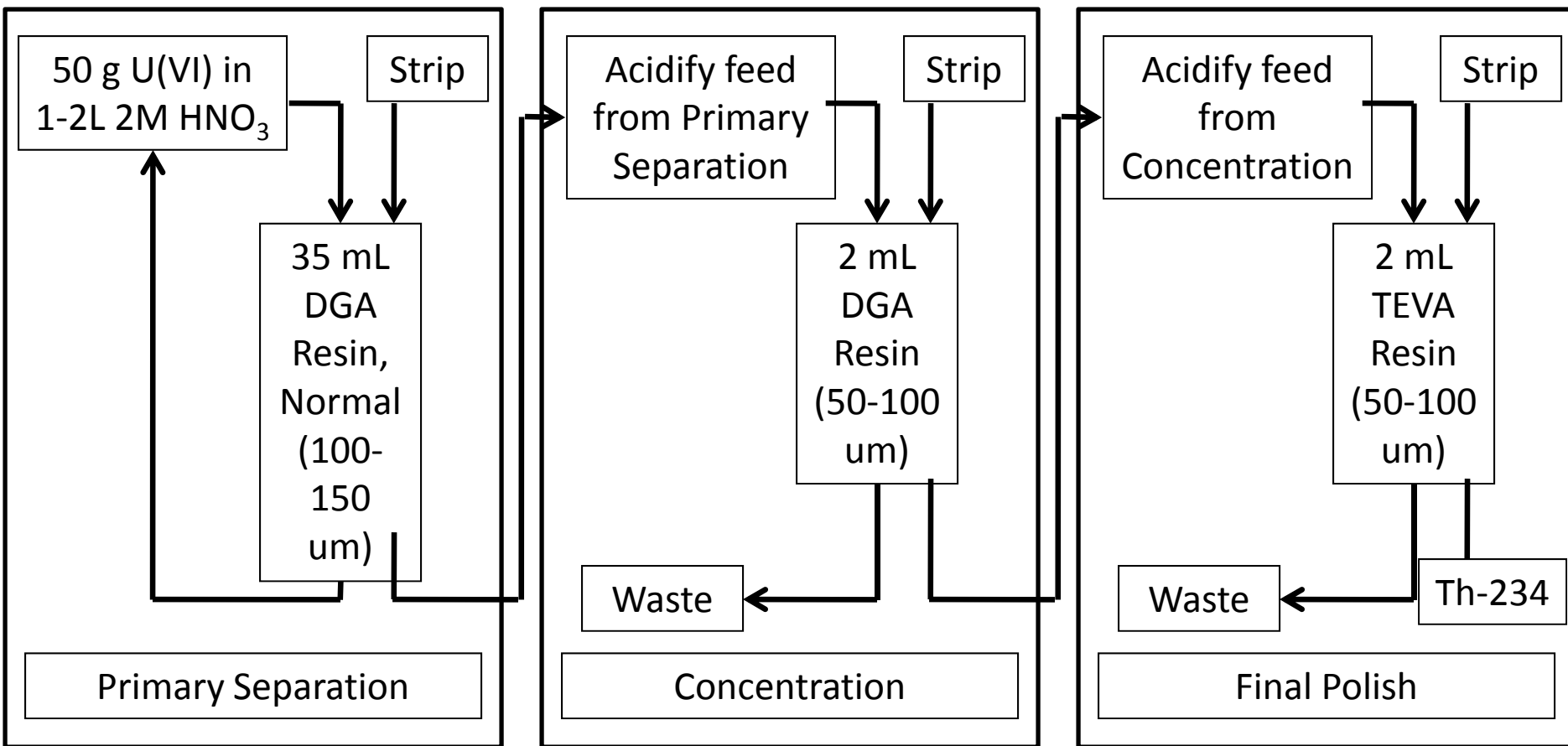


^{234}U 2.45×10^5 years

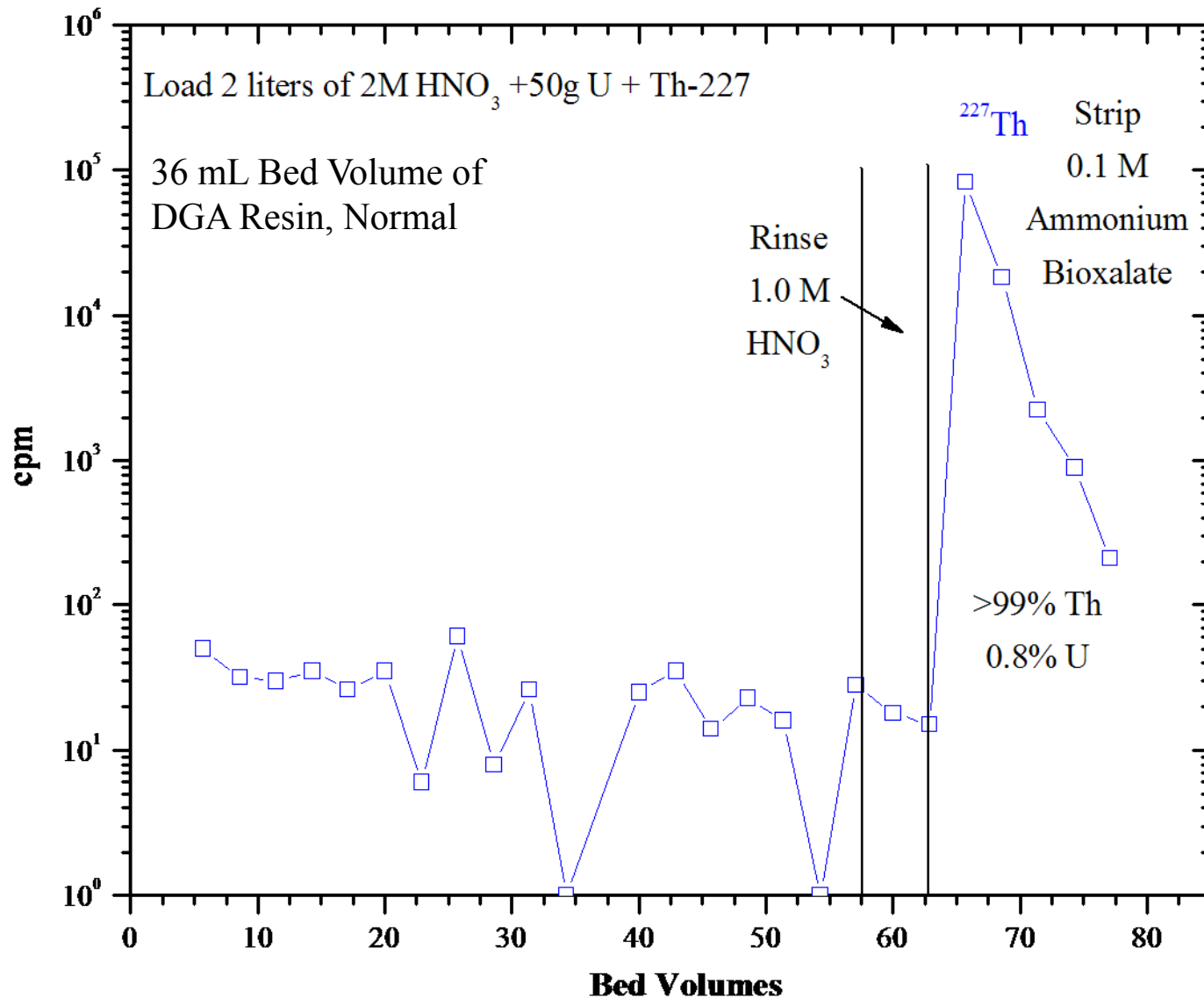


^{230}Th 8.0×10^4 years

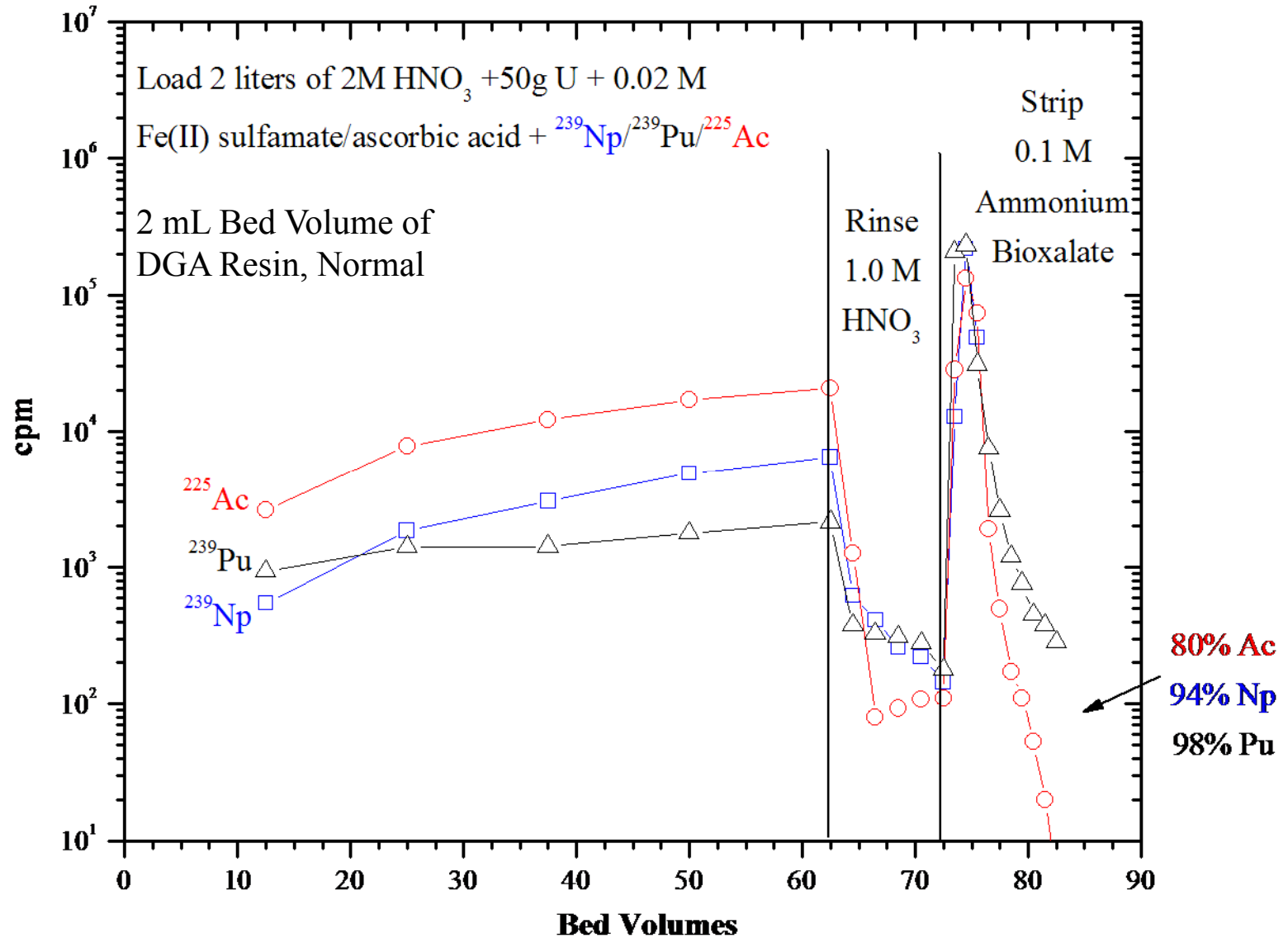
^{234}Th Separation from 50 g ^{238}U



Primary Separation



Separation of Np, Ac, Pu from 0.1 M U



Decay of ^{233}U

^{233}U 1.59 x 10⁵ years

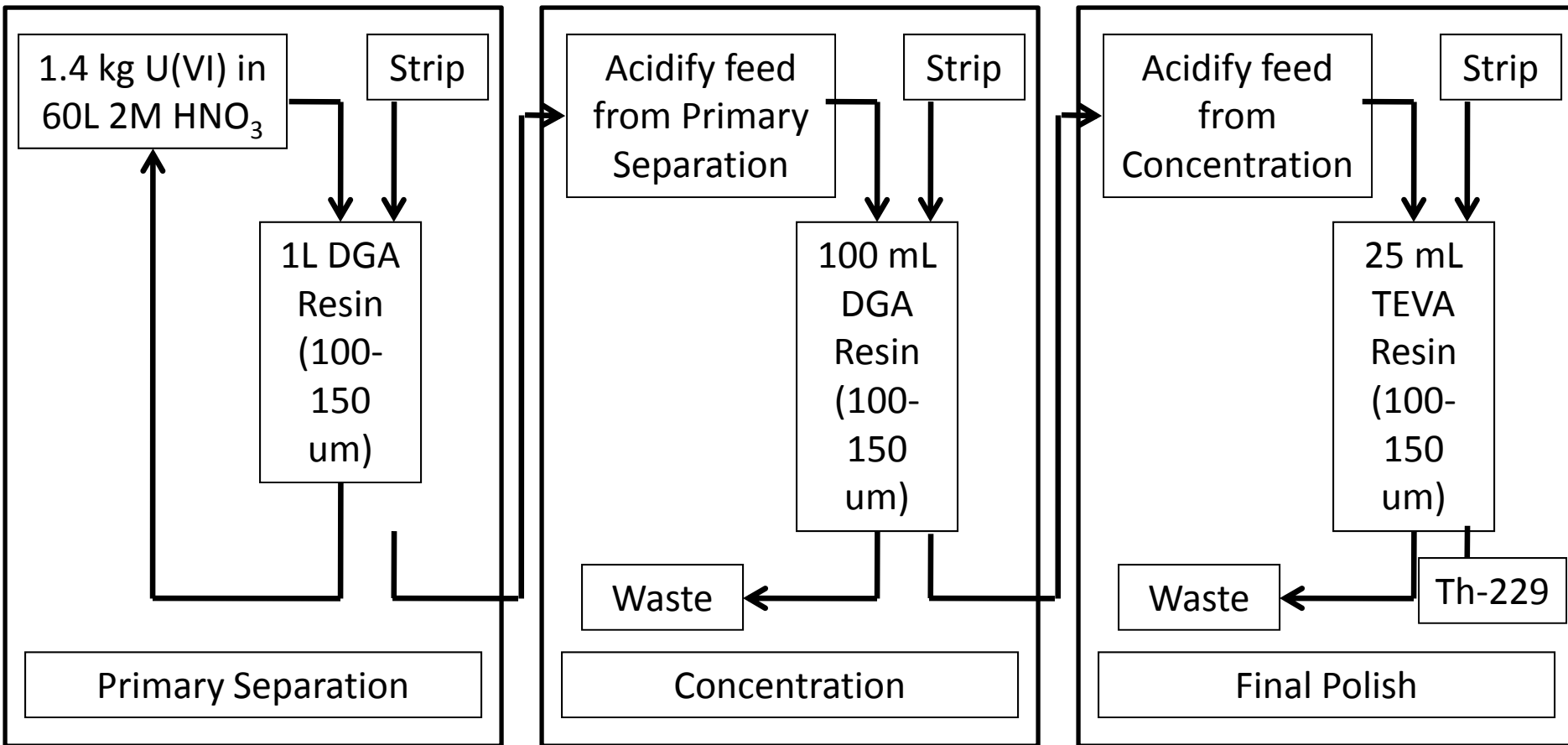


^{229}Th 7.3 x 10³ years

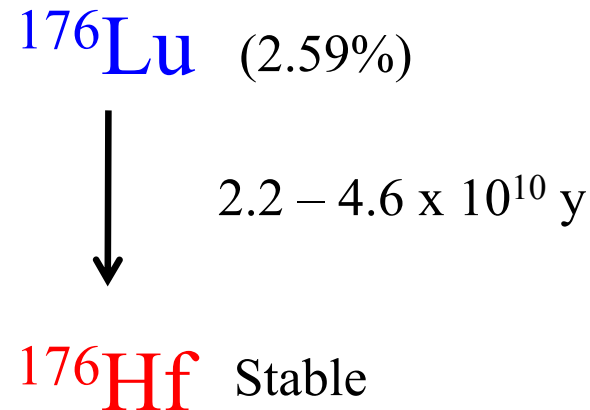


^{225}Ra 14.8 days \longrightarrow ^{225}Ac 10.0 days

^{229}Th Separation from 1.4 kg ^{233}U

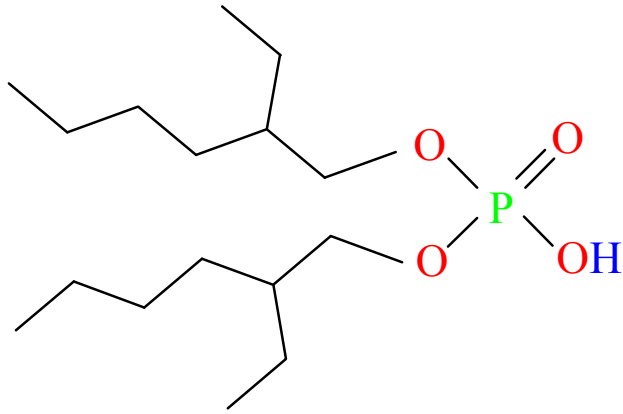


Separation of ^{176}Hf from kg amounts of Lu

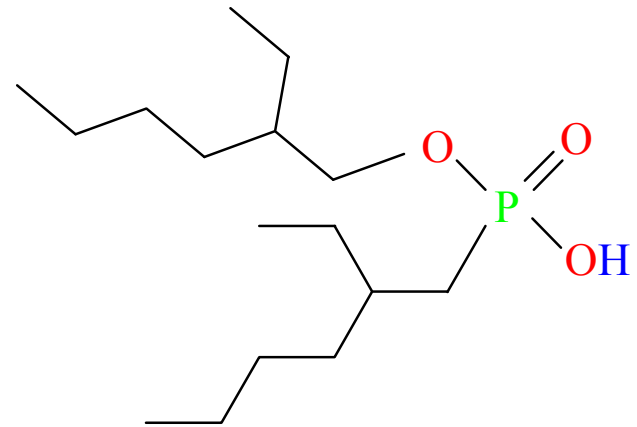


$$^{176}\text{Lu } t_{1/2} = \begin{array}{l} 3.8 \times 10^{10} \text{ y (CRC - 2003)} \\ 3.59 \times 10^{10} \text{ y (Browne/Firestone - 1986)} \end{array}$$

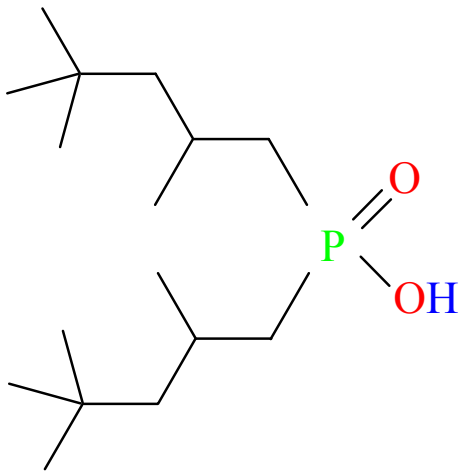
The LN Series of Resins



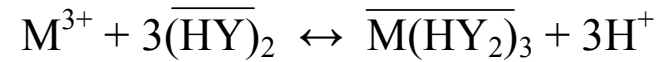
HDEHP (LN)



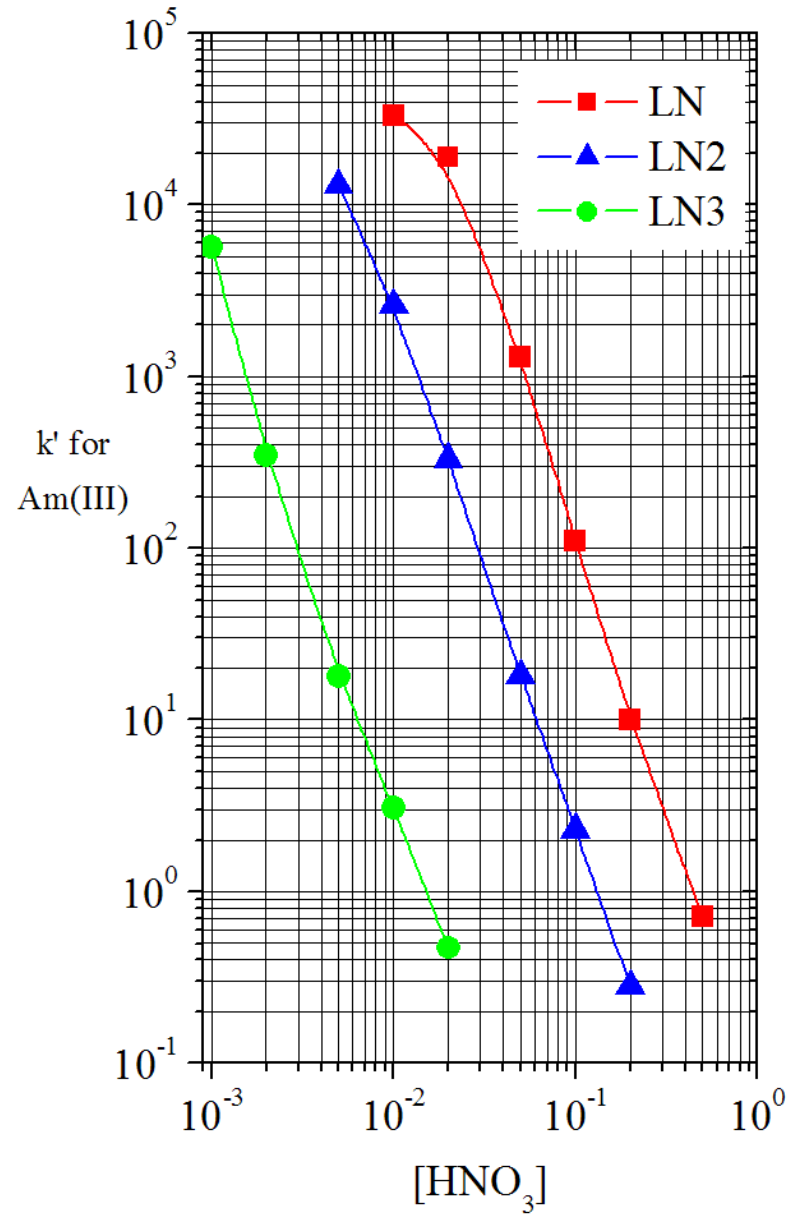
HEH[EHP] (LN2)



H[DTMPPeP] (LN3)

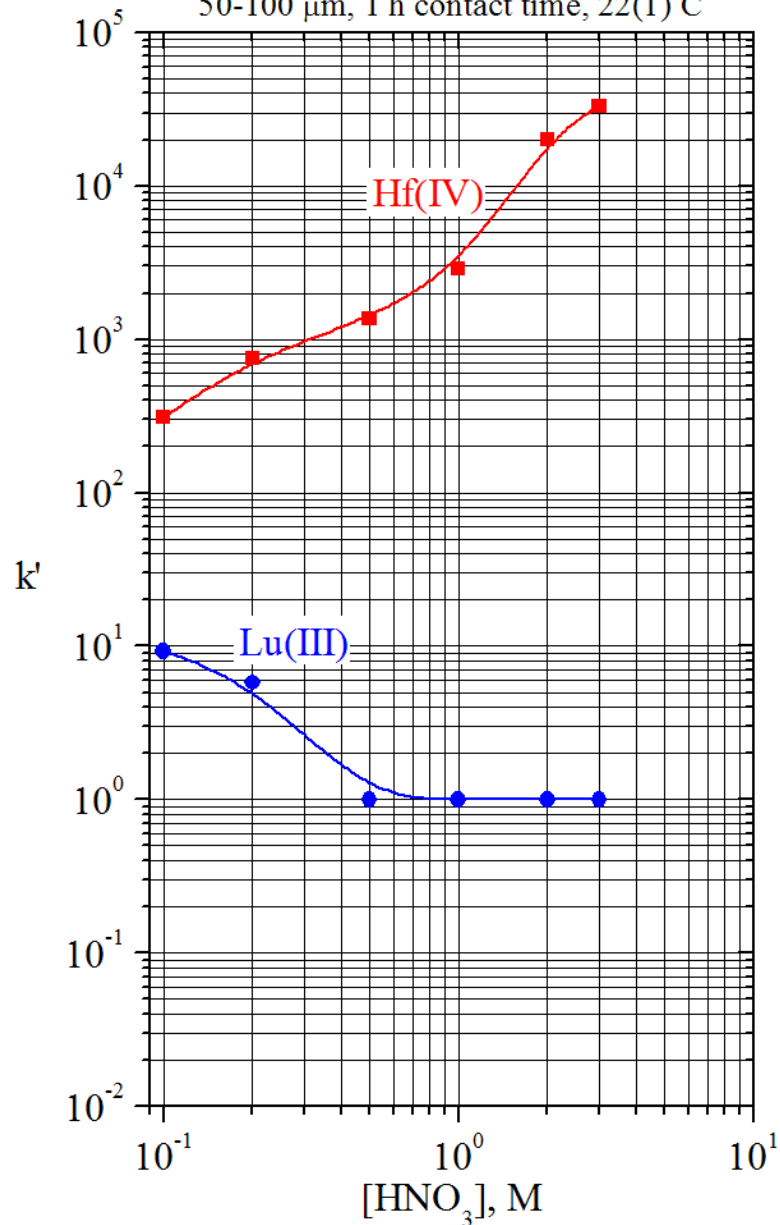


k' Am(III) on LN, LN2 and LN3 vs HNO₃
50-100 μm, 1 hr equilibration, 22(1)°C



k' Hf and Lu on Ln3 Resin vs. HNO₃

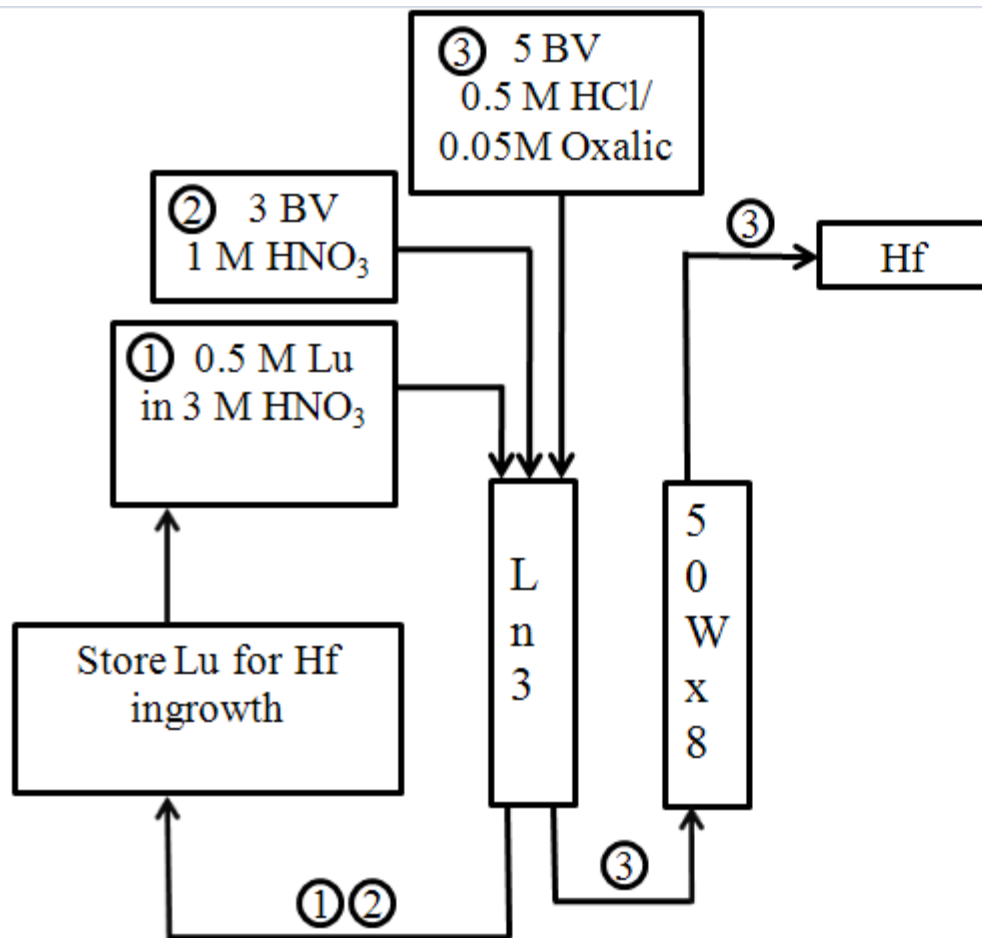
50-100 μm, 1 h contact time, 22(1)°C



$\alpha_{\text{Hf/Lu}}$ on Ln-3 Resin

HNO ₃	k'-Hf	k'-Lu	$\alpha_{\text{Hf/Lu}}$
0.1	312	9.3	34
0.2	753	5.8	130
0.5	1365	<1	>1300
1	2891	<1	>2900
2	20164	<1	>20,000
3	33245	<1	>33,000

Separation of ^{176}Hf from kg amounts of Lu



$\text{Ln } 3 = \sim 25 \text{ mL/L } 0.5\text{M Lu in } 1\text{M HNO}_3$

$50\text{wx}8 = 1 \text{ mL}/10 \text{ mL Ln}3$

Precondition Ln3 with 1 Bed Volume of 1M HNO₃

Precondition 50Wx8 with 2 Bed Volumes of 0.5M HCl/0.05M Oxalic acid

2.0 mL cartridge Ln3, 22(1)°C, 50-100 μm, 2 mL/min

